

High Luminosity Polarized ^3He Targets For Electron Scattering Experiments

Advisor: Prof. Gordon Cates



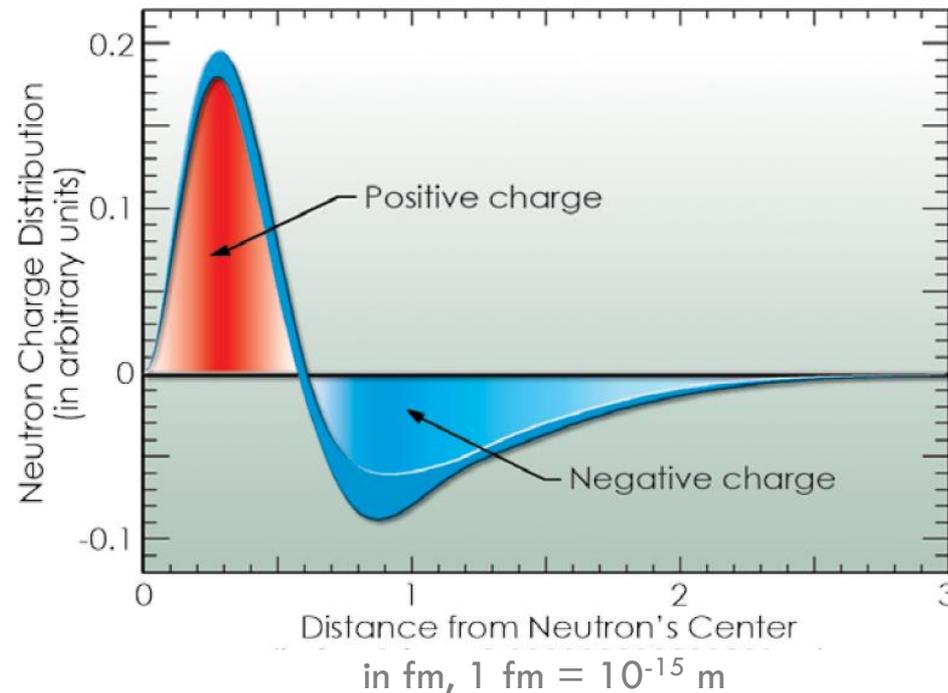
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University of Virginia

5/01/2014

Elastic form factors of the neutron

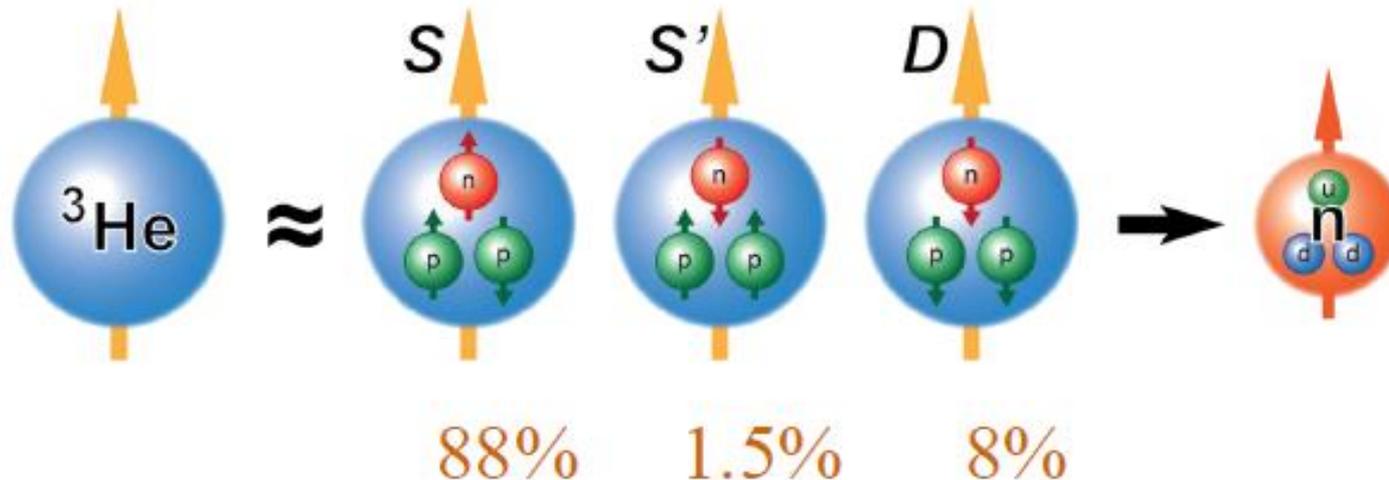
Among the reasons they are important,

- A fundamental property of the neutron
- Provides information on the charge density inside neutron
- Tests models of neutron structure
- Provides critical constraints on generalized parton distributions (GPD's)



Neutron Target

- Neutron mean lifetime is just under 15 mins.
- ^3He nucleus has two protons whose spins are paired, and a single neutron that accounts for most of the nuclear spin.
- So, ^3He is an effective polarized neutron target.



Effective Luminosity

In a double-spin asymmetry experiment

$$A_{measured} = P_e P_{He} A_{physical}$$

Relative error of Asymmetry is given by,

$$\frac{\delta A_{physical}}{A_{physical}} = \frac{1}{\sqrt{N} P_e P_{He}}$$

$$N \propto Lt$$

$$L = I_{beam} [\text{He}] l$$

$$\left(\frac{\delta A_{physical}}{A_{physical}} \right)^2 \propto \frac{1}{L P_{He}^2}$$

 Effective Luminosity

The performance of polarized ^3He targets has increased by roughly a factor of 30 since SLAC E142

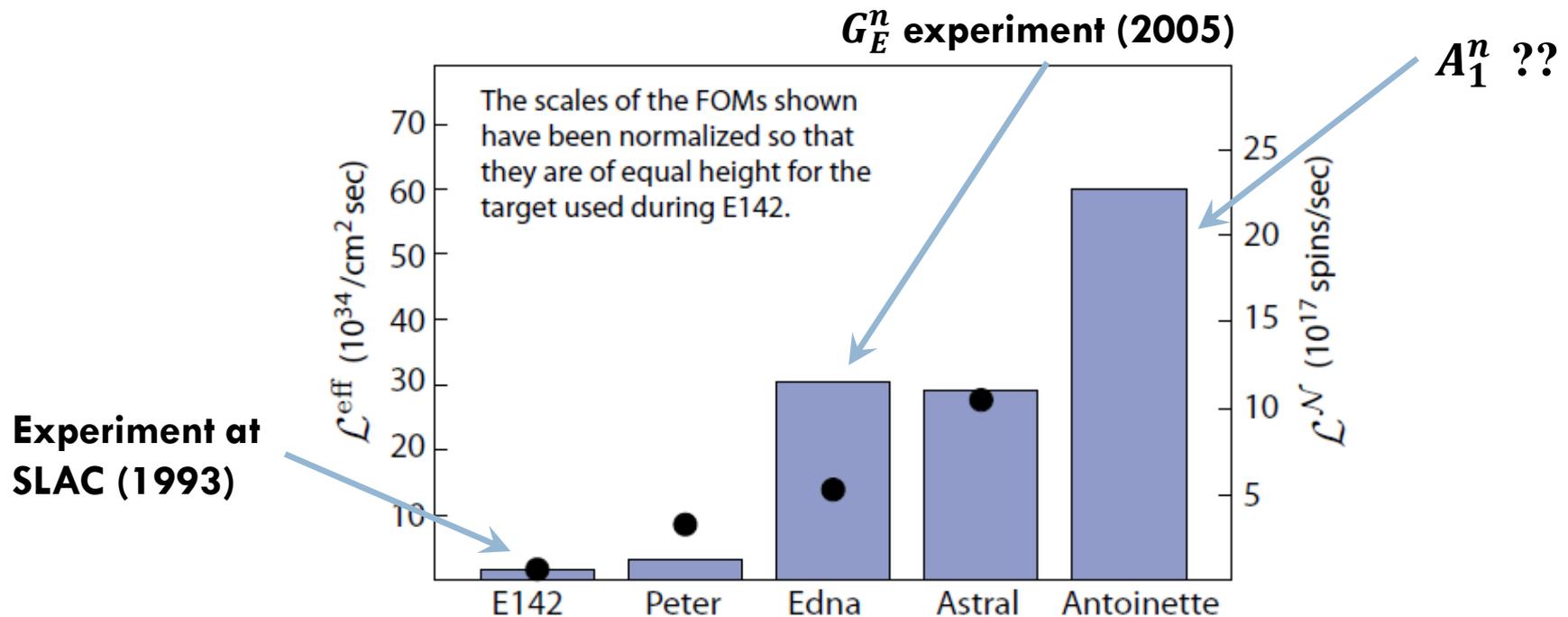
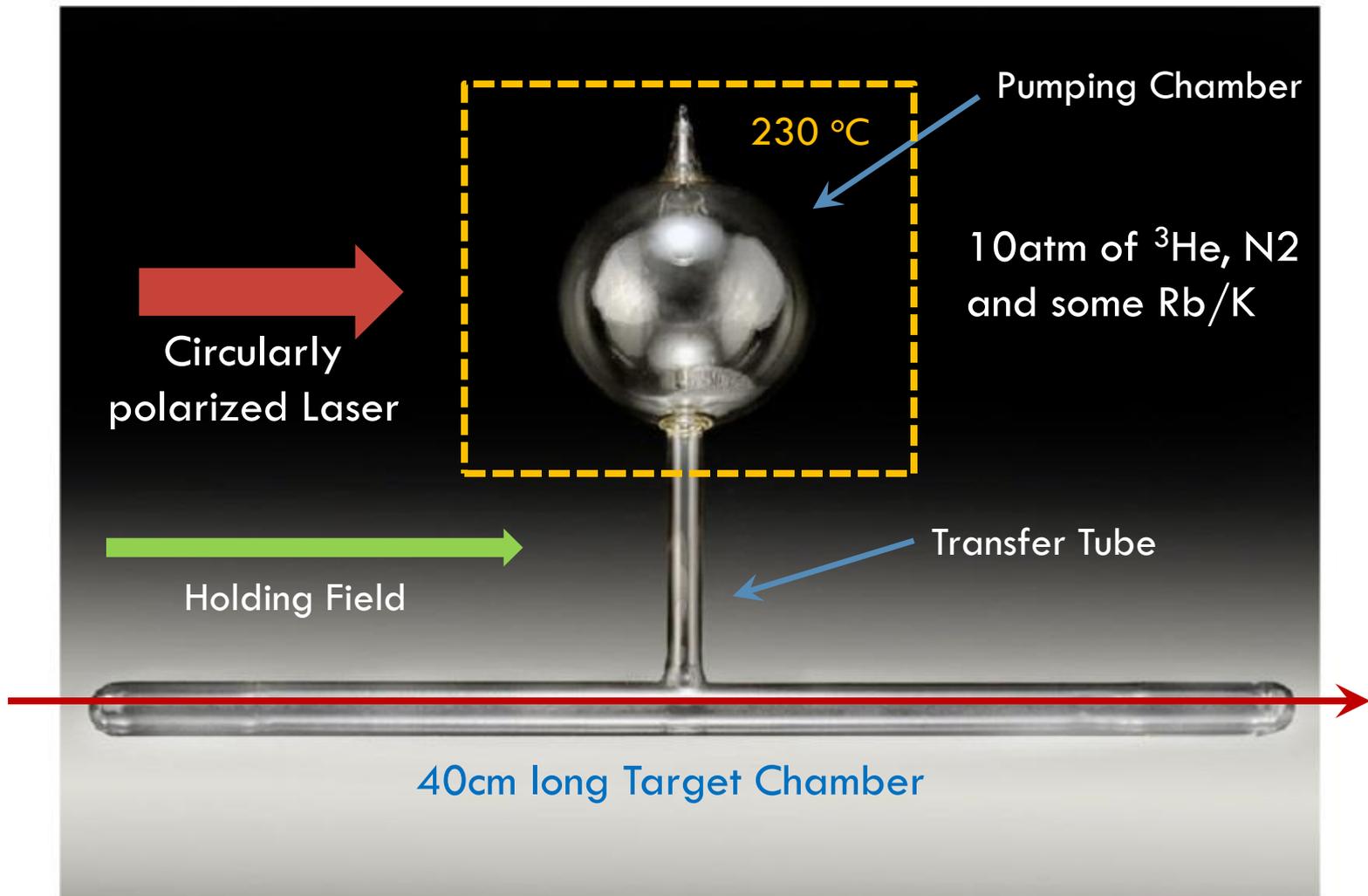


FIG. 1: Shown are two figures of merit (FOM) for five polarized ^3He targets. The solid circles (left-hand scale) indicate the luminosity weighted by ^3He polarization squared (P_{He}^2) achieved in beam. The shaded columns (right-hand scale) show a FOM proportional to the total number of spins polarized per second, again weighted by P_{He}^2 .

^3He target cell

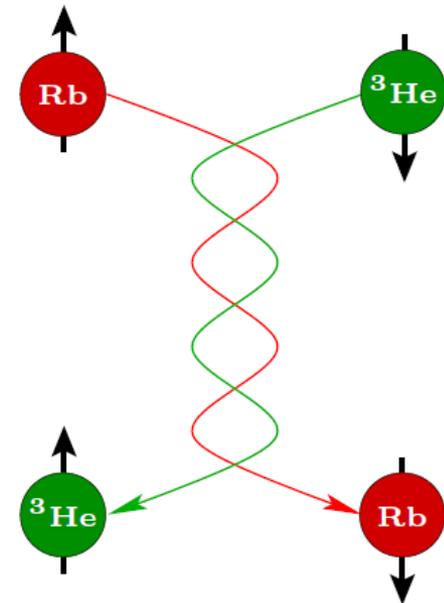
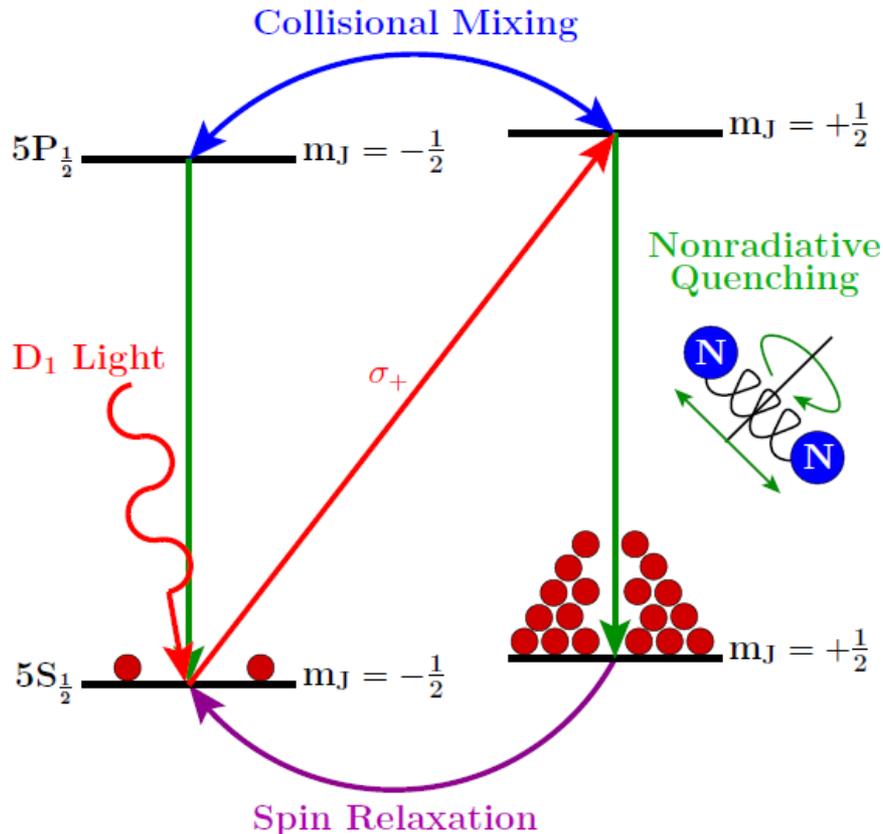


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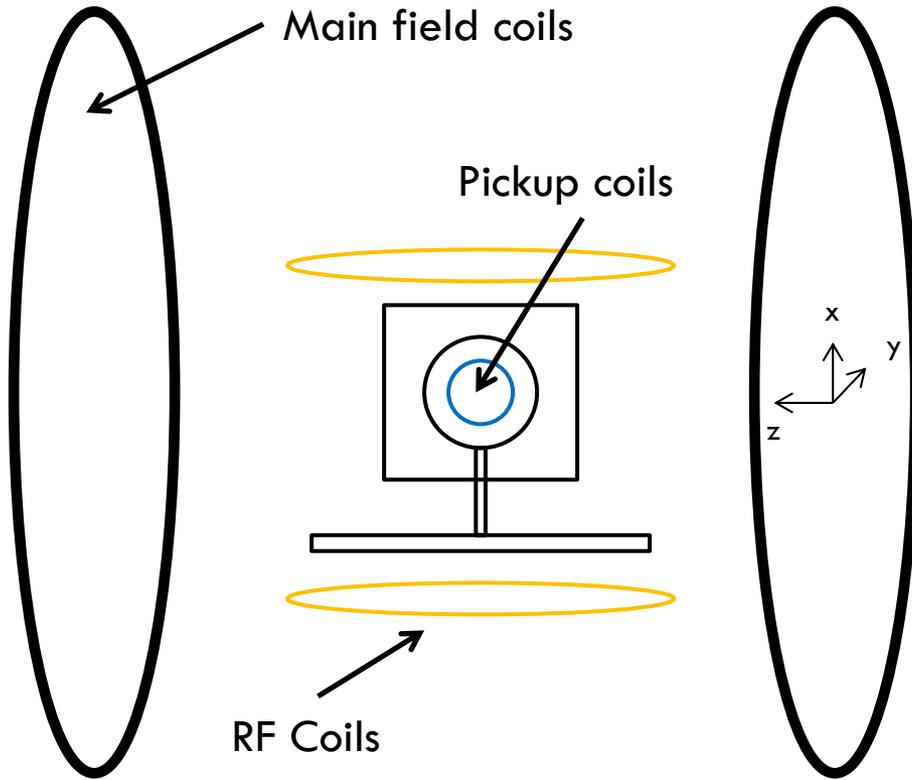
- Techniques
- Convection based Cells
- Targets with metal target windows

Spin Exchange Optical Pumping

- Use Spin Exchange Optical Pumping (SEOP) method to polarize the target.
 1. an alkali-metal vapor is polarized by optical pumping using lasers.
 2. the polarized alkali-metal atoms transfer spins to ^3He atoms.



Adiabatic Fast Passage (AFP)



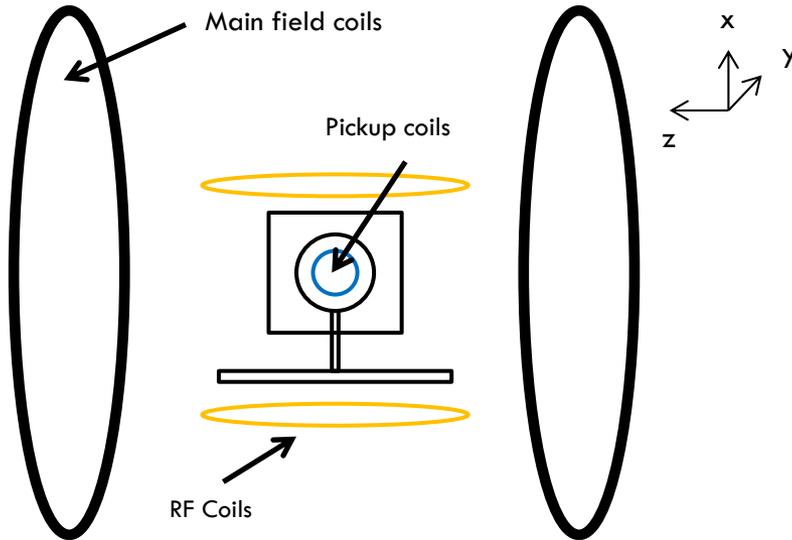
$$\vec{B}_0 = B_0 \hat{z}$$

$$\vec{B}_1 = B_1 \cos \omega t \hat{x}$$

In rotating coordinates

$$B = (B_0 - \omega/\gamma) \hat{z} + B_1 \hat{x}$$

Adiabatic Fast Passage (AFP)

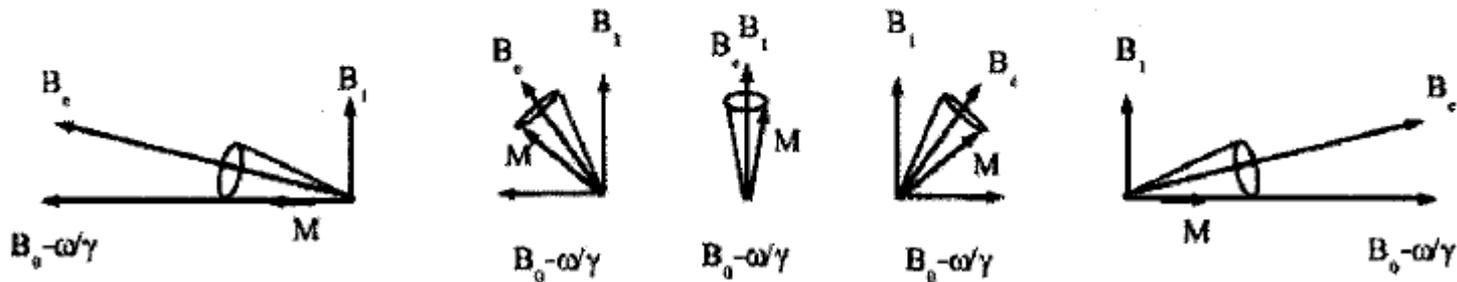


$$\vec{B}_0 = B_0 \hat{z}$$

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In rotating coordinates

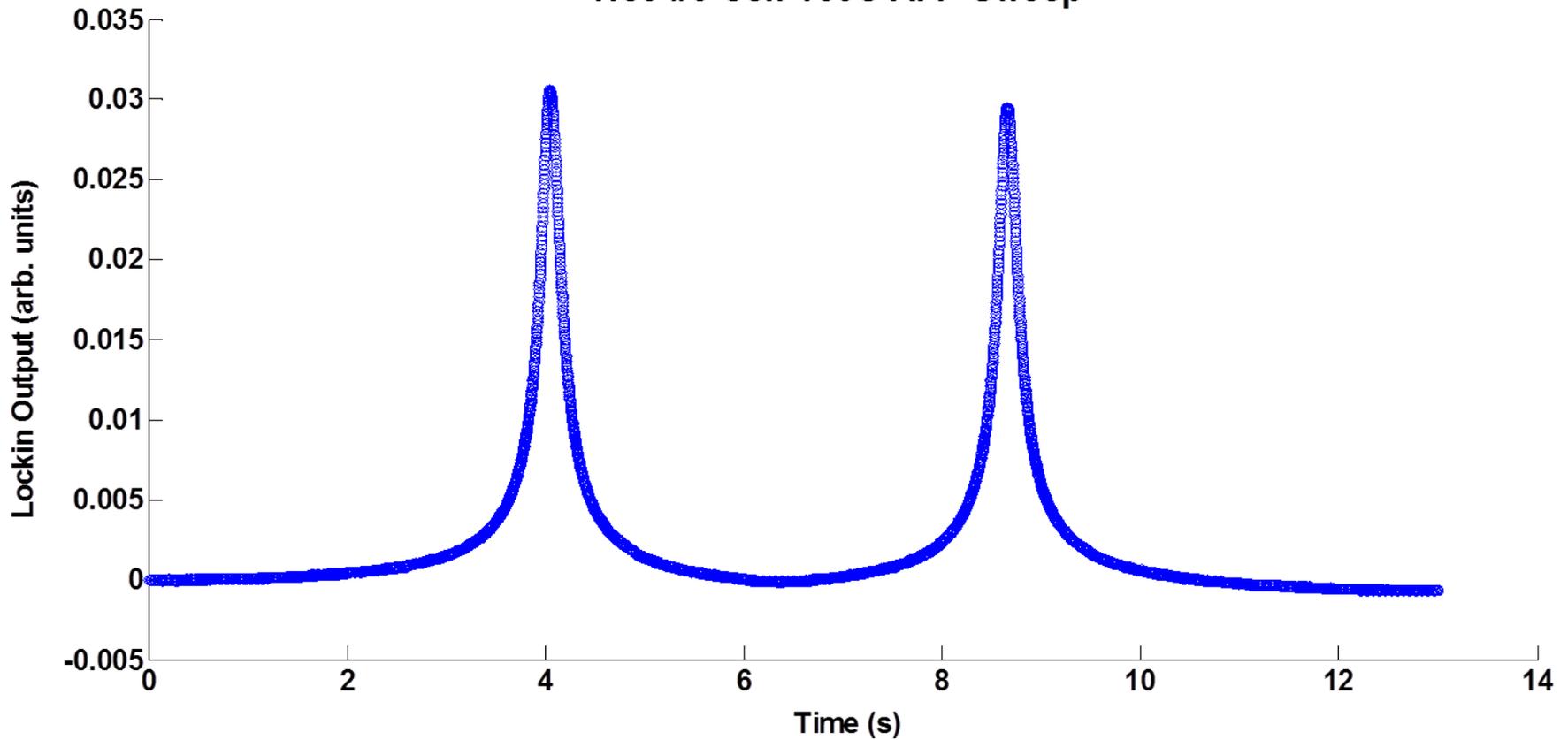
$$B = (B_0 - \omega/\gamma) \hat{z} + B_1 \hat{x}$$



- Net magnetization follows effective magnetic field.

AFP Signal

He3 #6 Cell 160C AFP Sweep



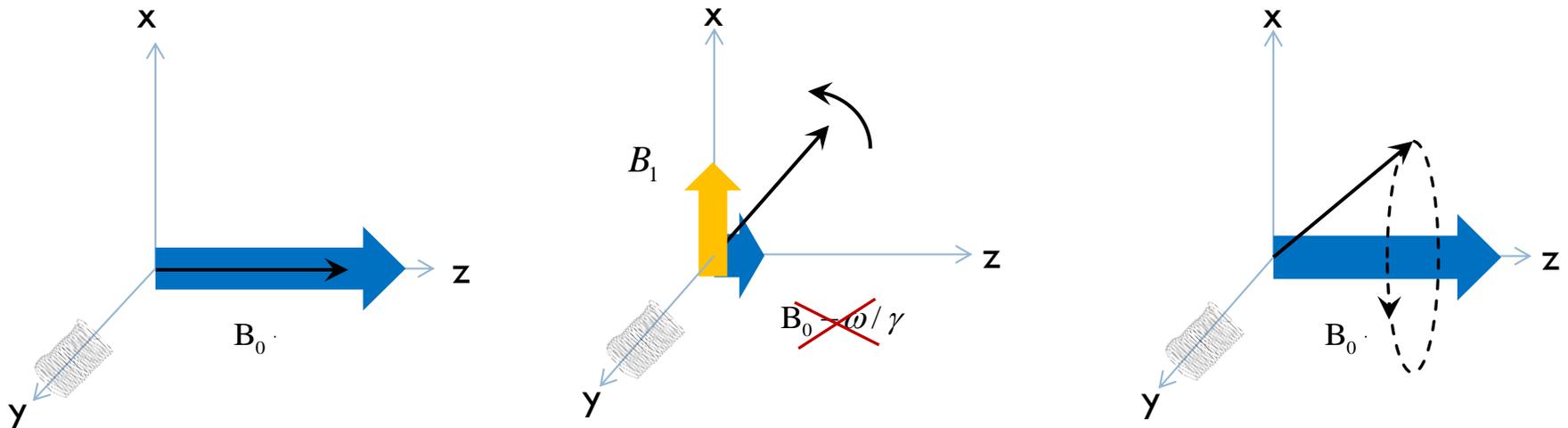
Pulse NMR

- Radio frequency energy is applied in the form of a short pulse on resonance which tips ^3He spins and make them precess around holding field.

In rotating coordinates

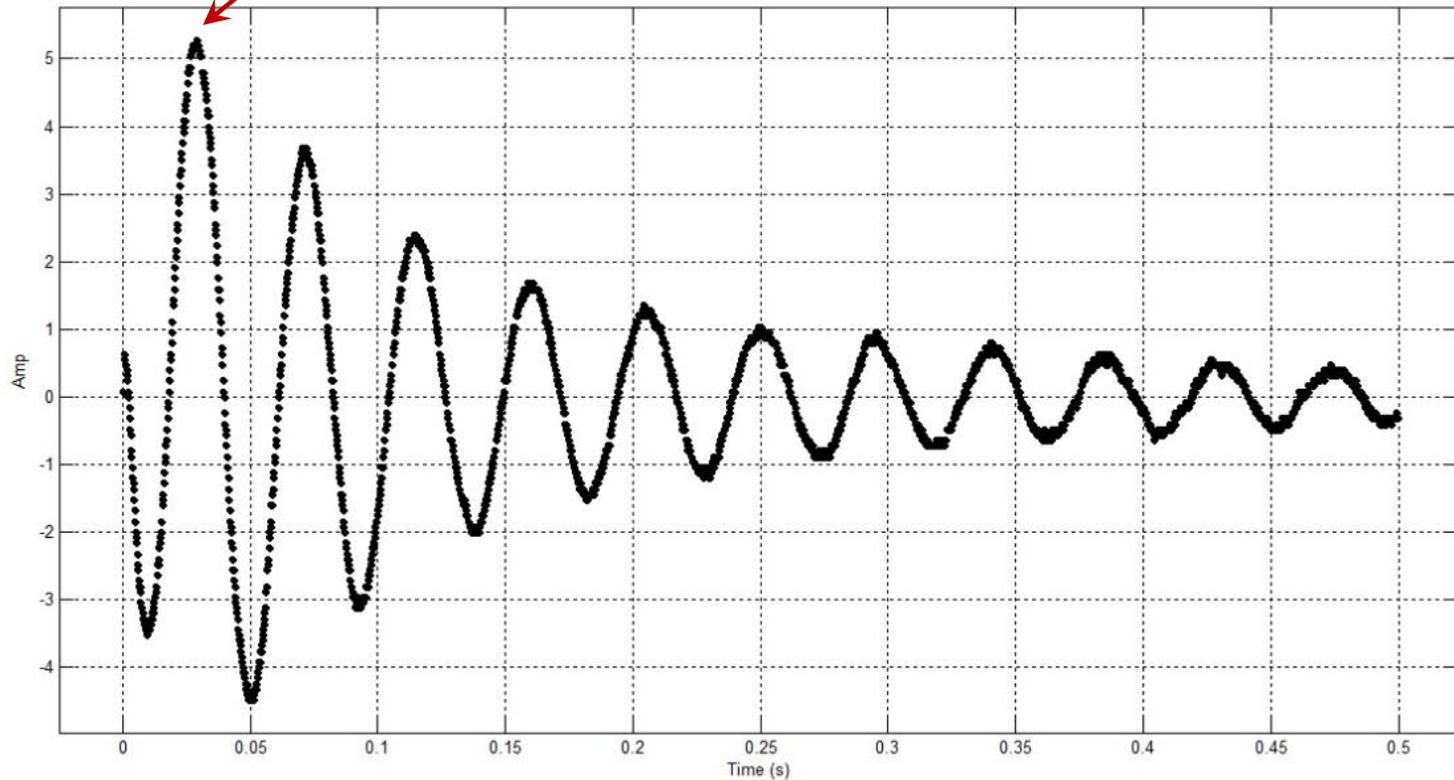
$$B = (B_0 - \omega/\gamma)\hat{z} + B_1\hat{x}$$

$$\vec{B}_1 = B_1 \cos \omega t \hat{x}$$



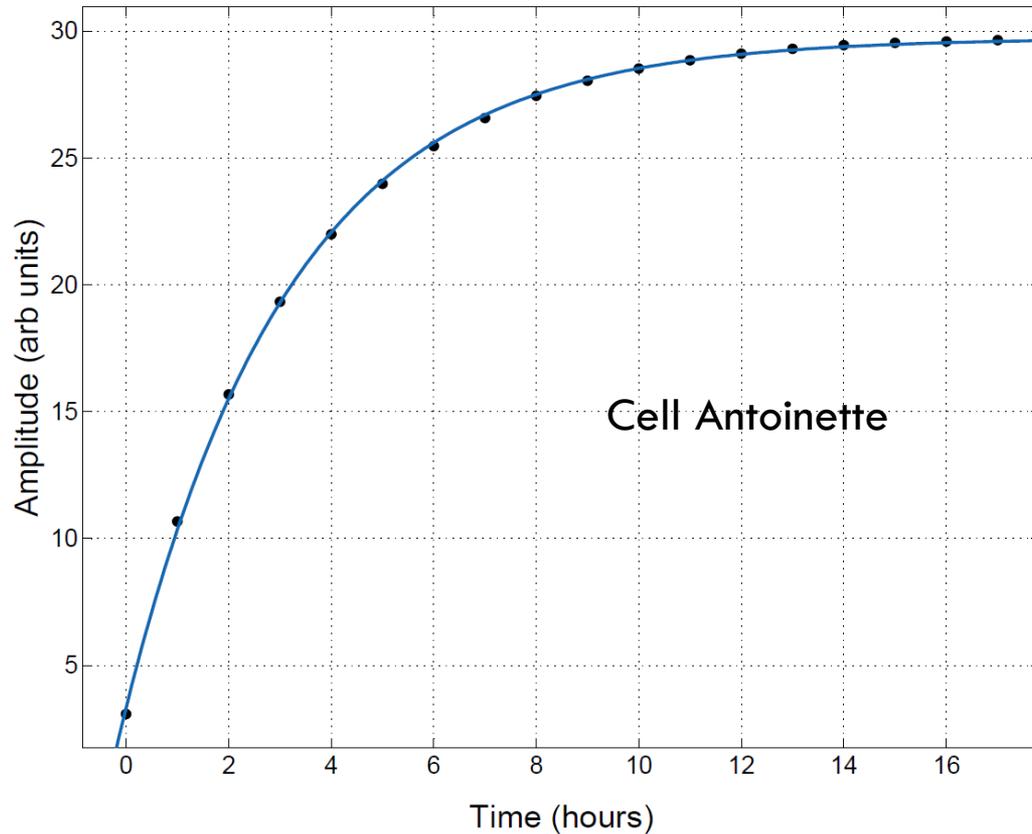
Pulse NMR Signal

Initial amplitude is taken as a measure of polarization



Spin Up Curve


Polarization



Polarization measured as a function of time, while cell is pumped up.

De-polarization or Spin Relaxation

Sources of ^3He spin relaxation $\Gamma = \Gamma_{dipolar} + \Gamma_{field} + \Gamma_{wall}$

□ Dipolar Relaxation

$$\Gamma_{dipolar} = \frac{[{}^3\text{He}]}{744} \text{hrs}^{-1}$$

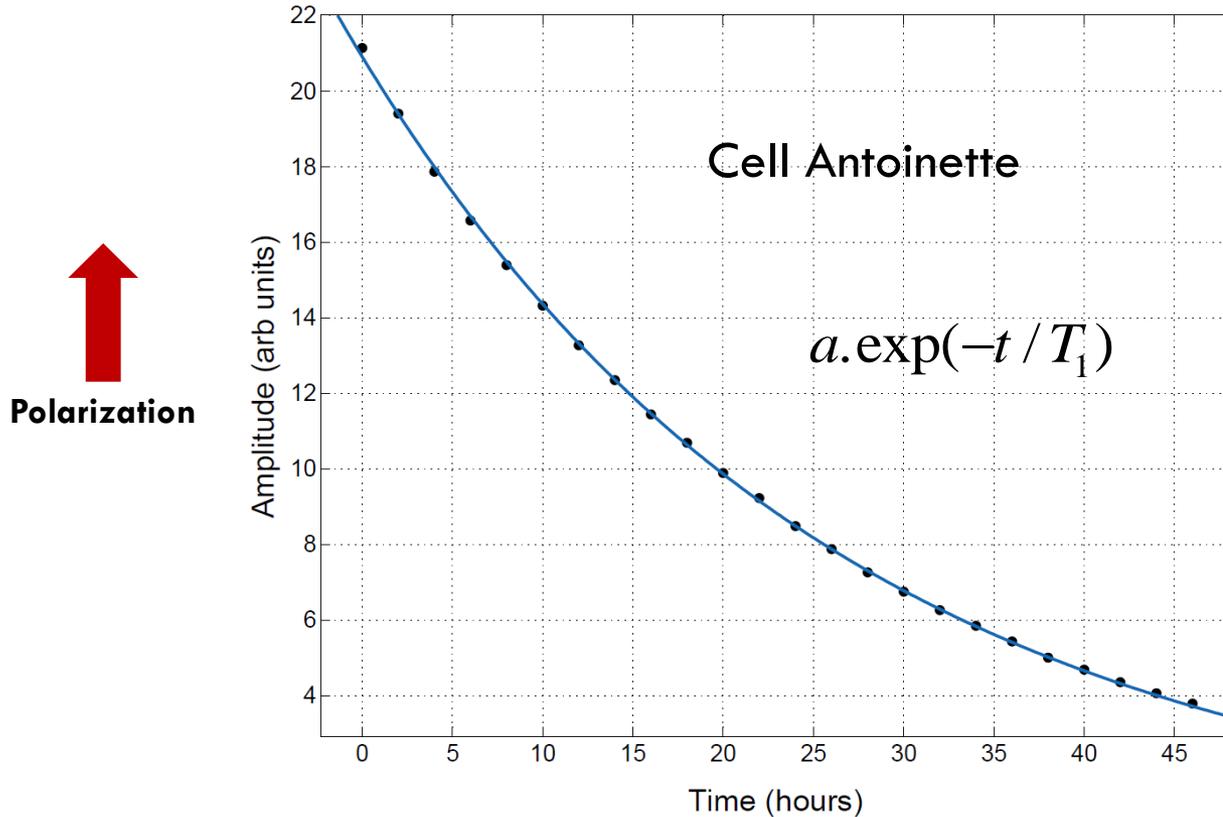
□ Static magnetic field inhomogeneity

$$\Gamma_{field} = D \frac{|\nabla B_x|^2 + |\nabla B_y|^2}{B_0^2}$$

□ Wall Relaxation

$$\Gamma_{wall} = \rho \frac{A}{V}$$

Spin Down Curve



$$\Gamma = \frac{1}{T_1}$$

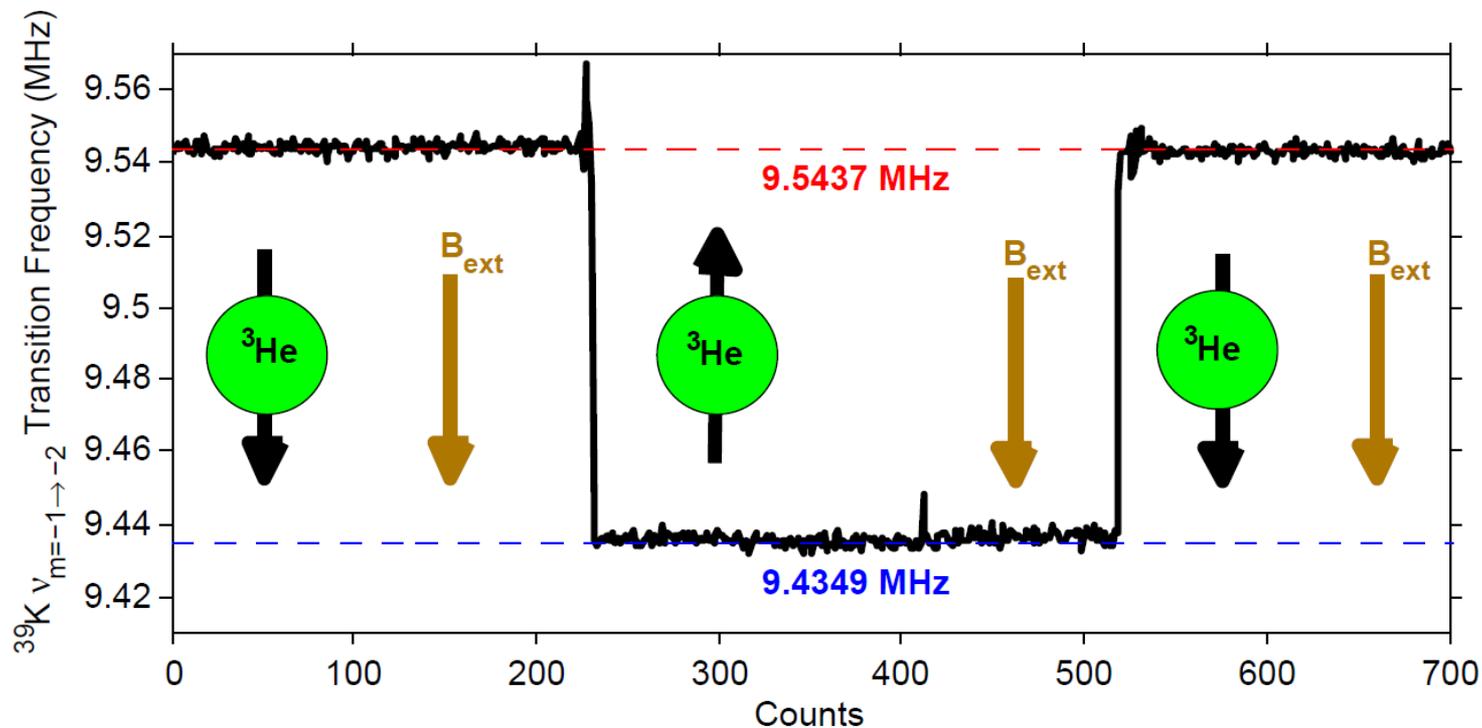
Polarization vs. time as spins relax.

Spin down measurement is used to determine life time of a cell.

Electron Paramagnetic Resonance (EPR)

- Rb/K Zeeman splitting is proportional to the size of the field.
- Alkali metal atoms experience a small “effective” field due to the presence of polarized ^3He gas.
- We can isolate this field due to the ^3He gas by flipping the ^3He spins.

Electron Paramagnetic Resonance (EPR)



Adapted from P.A.M. Dolph et al., 2011

$$\Delta\nu \propto P[\text{He}]$$

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Diffusion based Cell

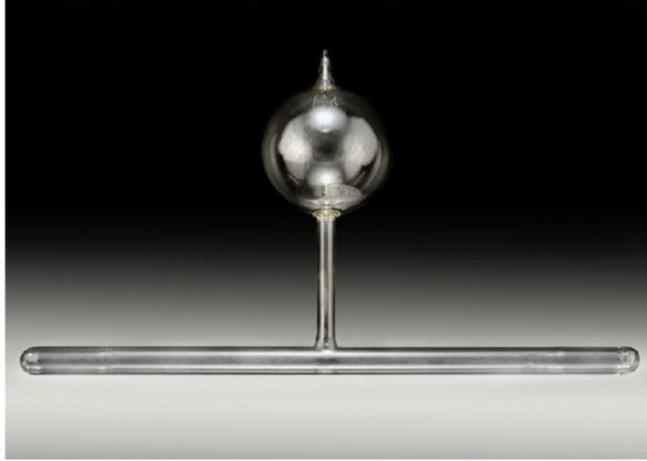


Photo Credit: A. Galyta

Historically used single transfer tube.

$$\frac{P_{tc}}{P_{pc}} = \frac{1}{1 + \Gamma_{tc}/d_{tc}}$$

Γ_{tc} - Spin relaxation rate in TC

d_{tc} - rate atoms leave TC

$$\Gamma_{tc}/d_{tc} \ll 1$$

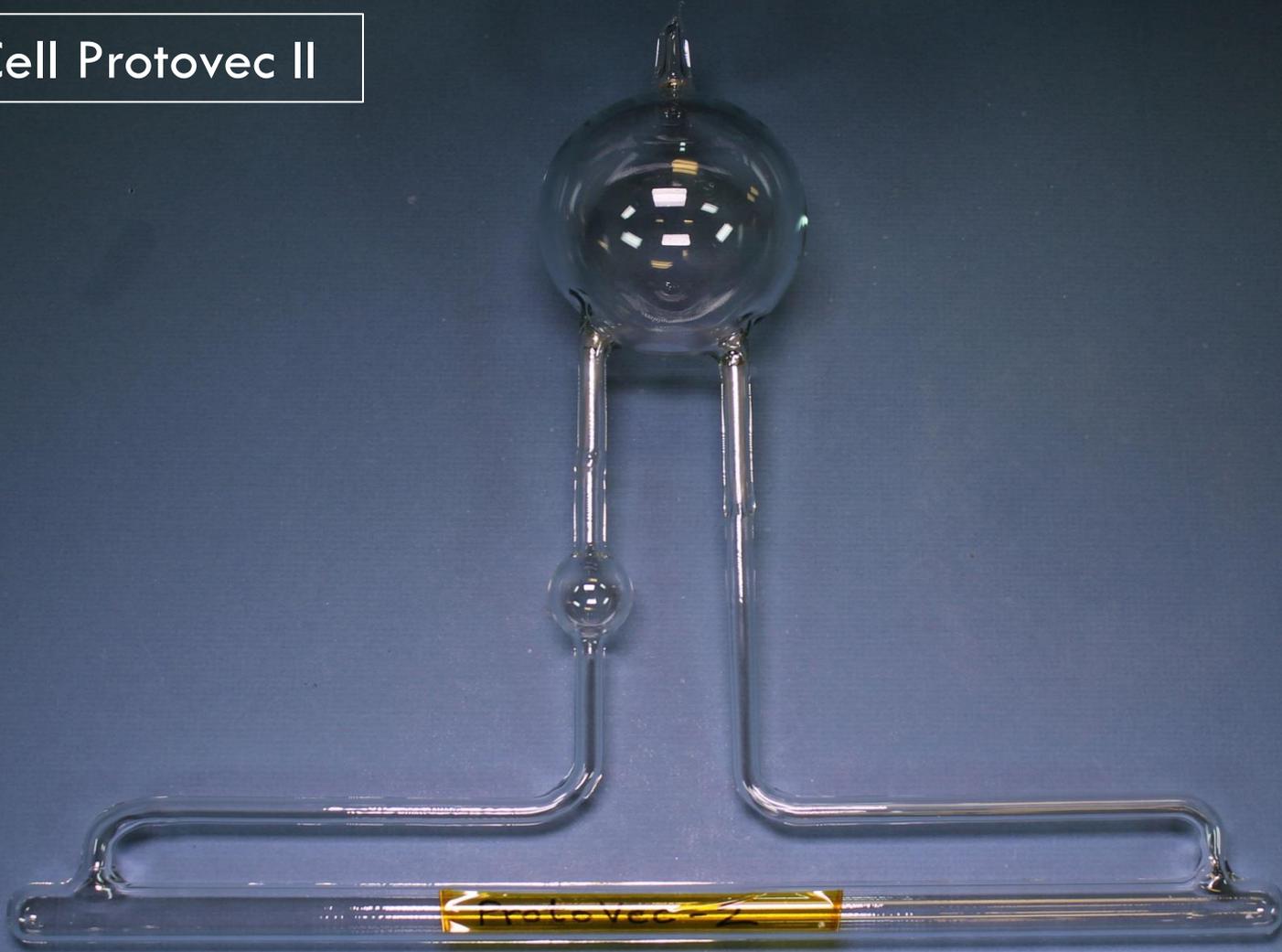
Difficult to maintain as beam current increases.

Results in lower target chamber in-beam polarization.

Since PC & TC are close to each other – risk for radiation damage.

Convection Cell

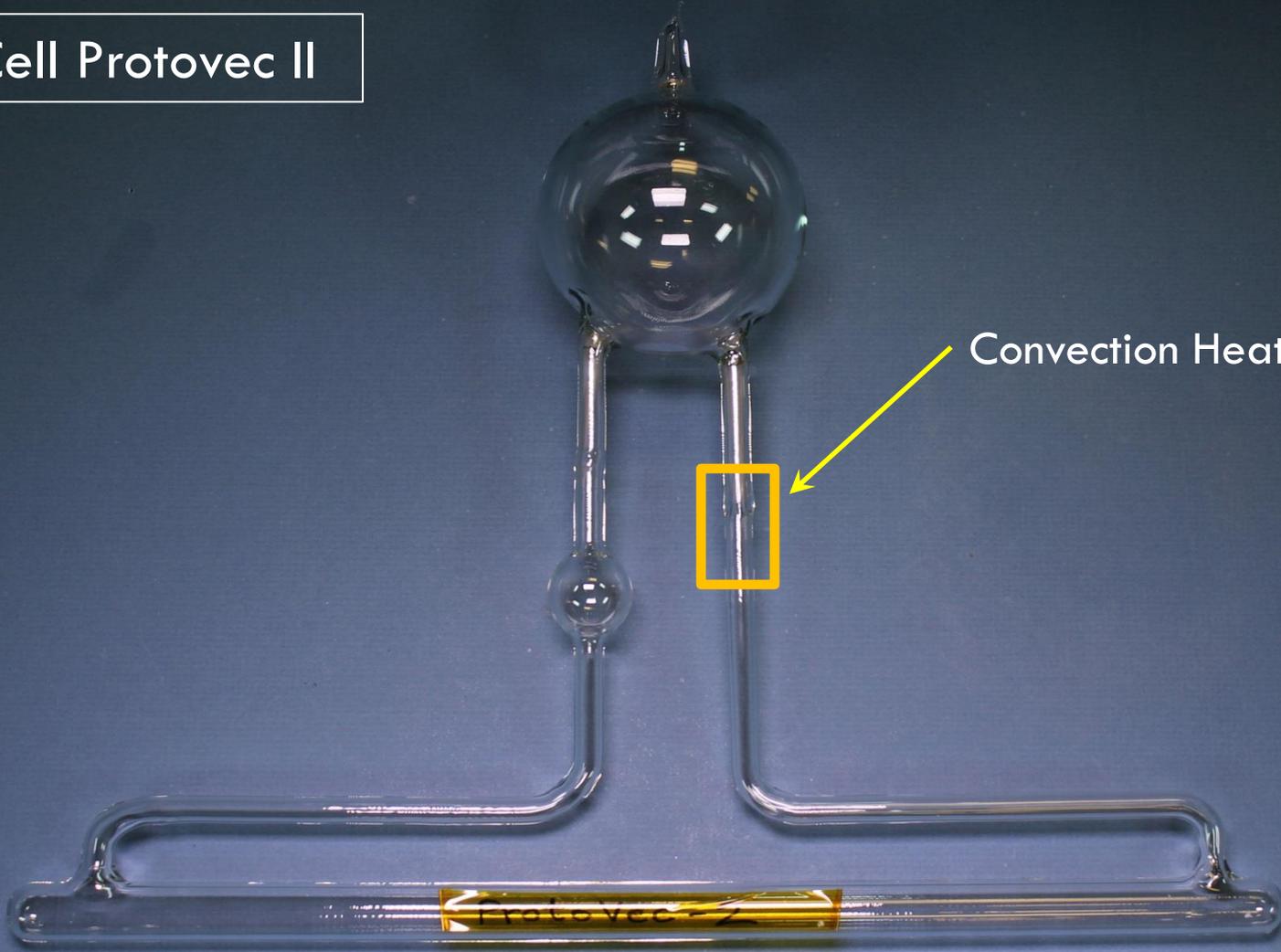
Cell Protovec II



Convection Cell

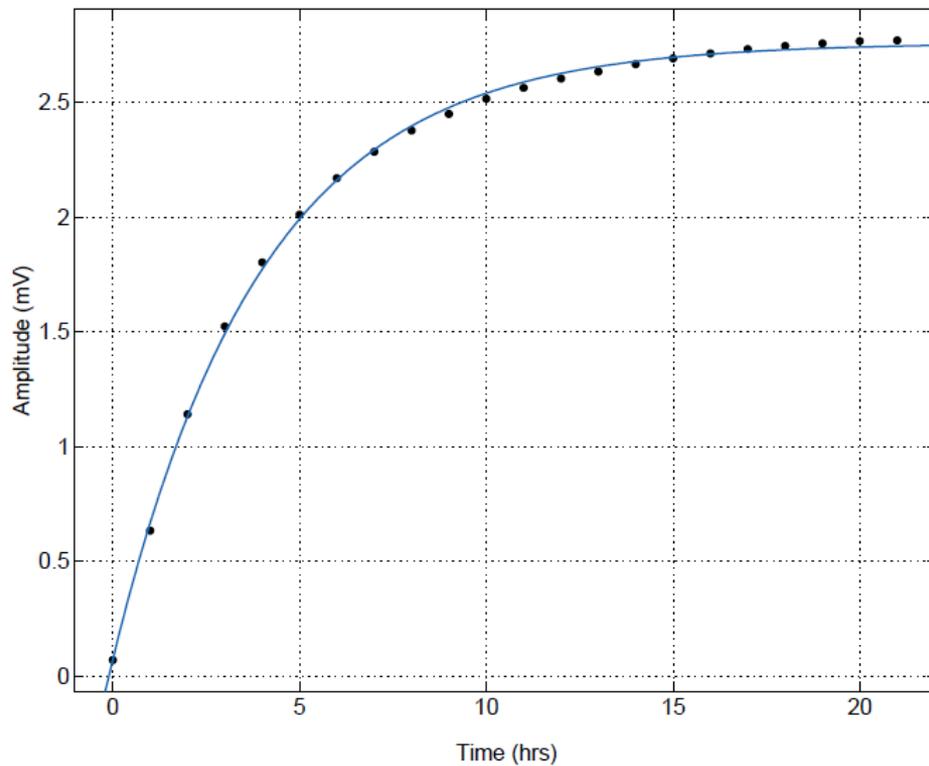
Cell Protovec II

Convection Heater

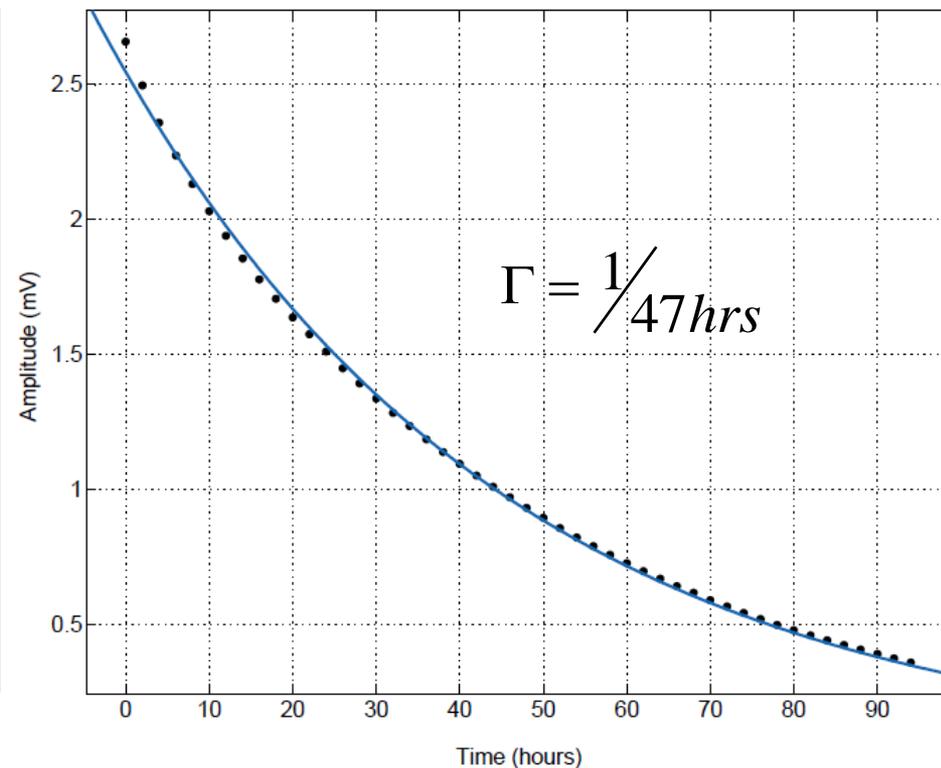


Protovec II Results

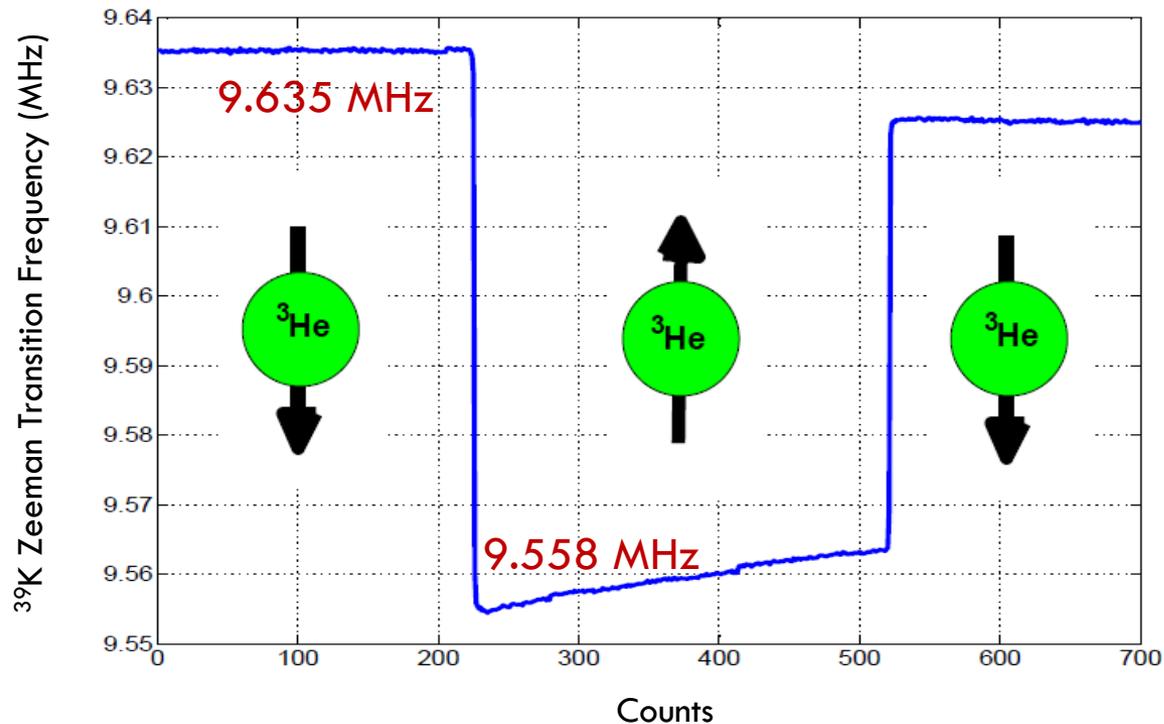
Spin Up at 235 °C



Spin Down



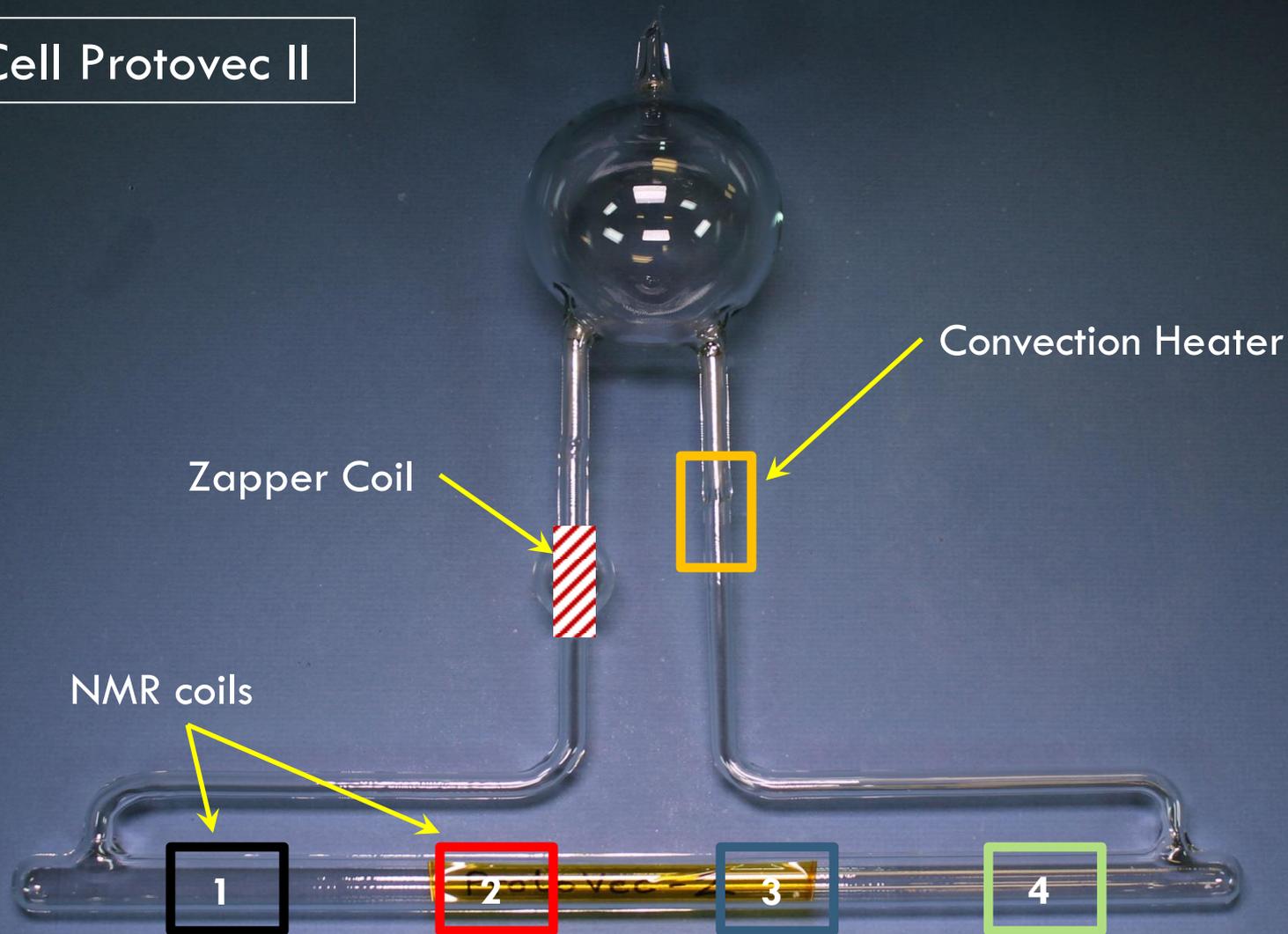
Protovec II Results



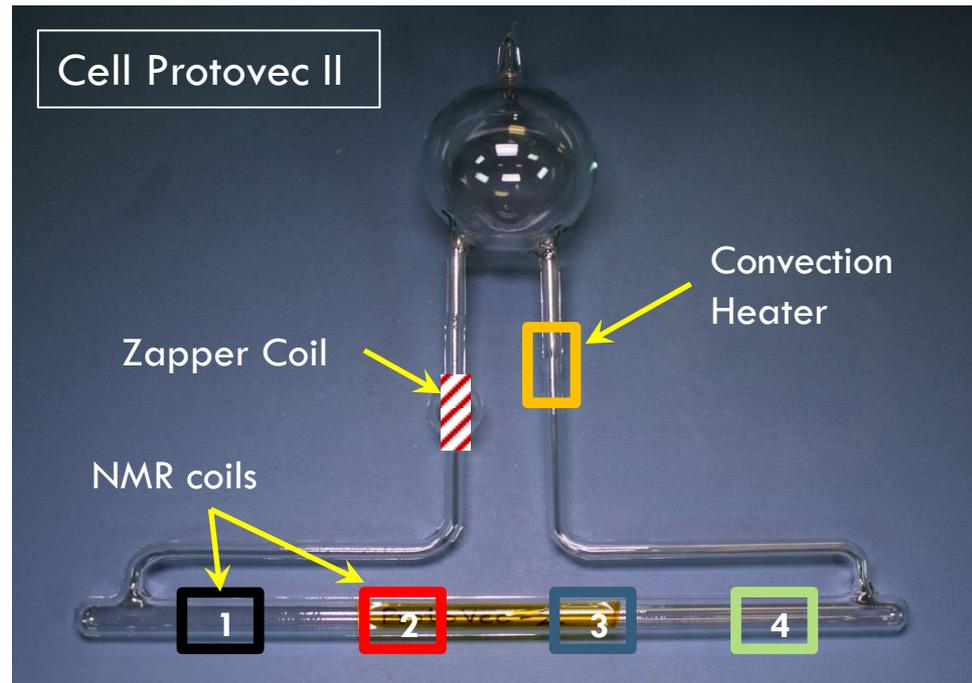
EPR calibration shows a $\sim 80\%$ saturated ^3He Polarization

Convection-driven gas-flow test

Cell Protovec II

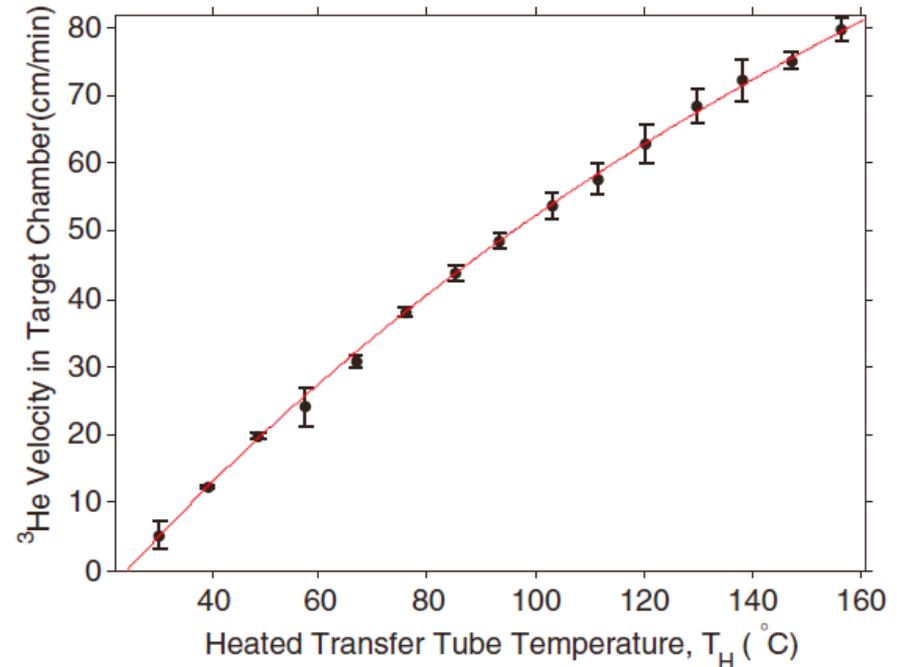
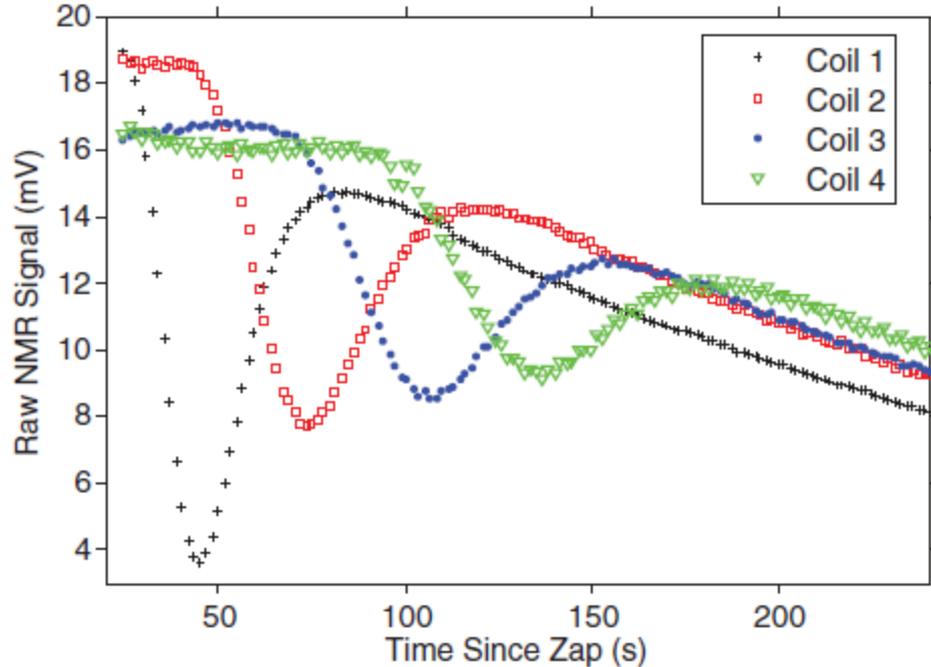


Convection-driven gas-flow test



- Heater drives convection
- Zapper coil depolarizes slug of gas
- Coils #1 - #4 monitor passage of depolarized slug.
- Heater temperature can be changed to adjust speed of gas.

Convection-driven gas-flow test



Adapted from P. A. M. Dolph et al., [Phys. Rev. C **84**, 065201 \(2011\)](#).

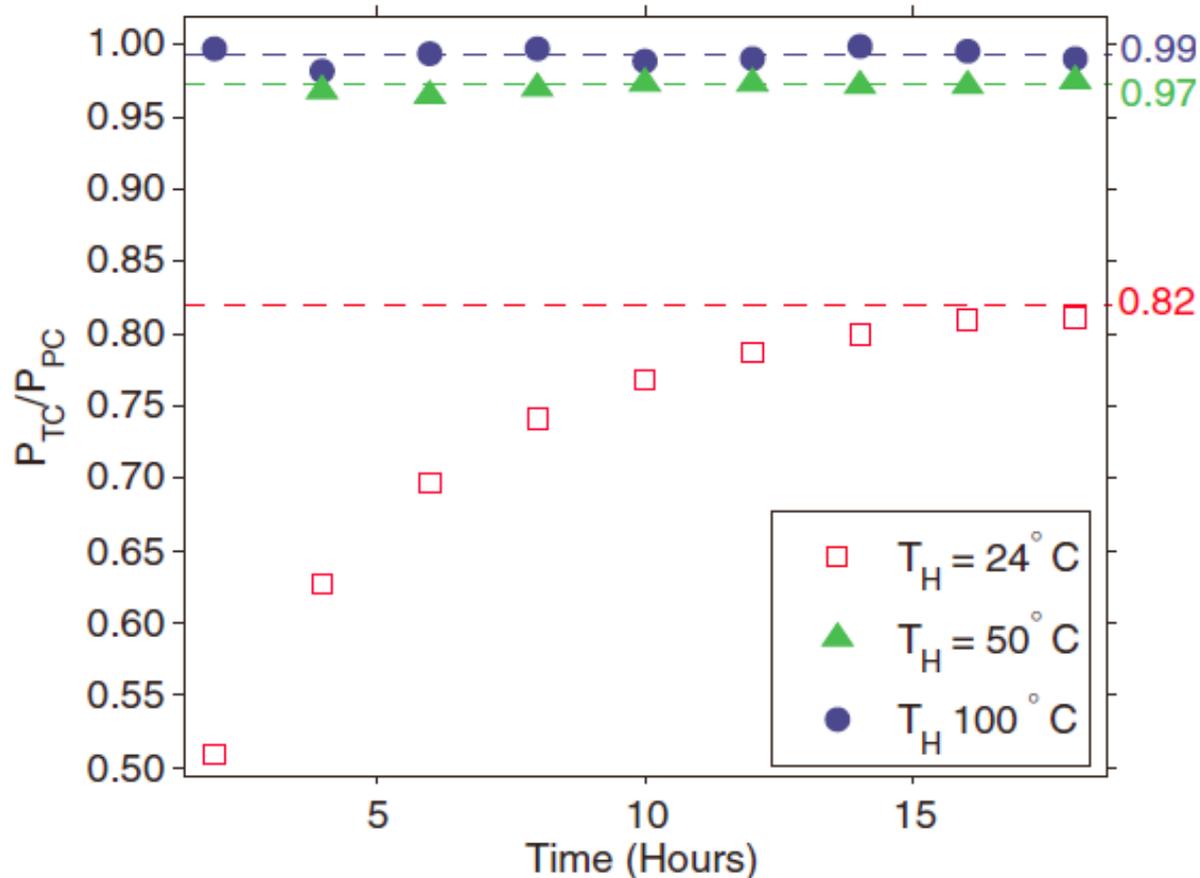
Unheated transfer tube was at 24 $^{\circ}\text{C}$.

The data indicate a gas flow velocity up to 80 cm/min.

d_{tc} for the diffusion based cell 0.72 h^{-1}

d_{tc} for the convection based cell 4.9 - 81 h^{-1}

Convection drives polarization gradient to zero.



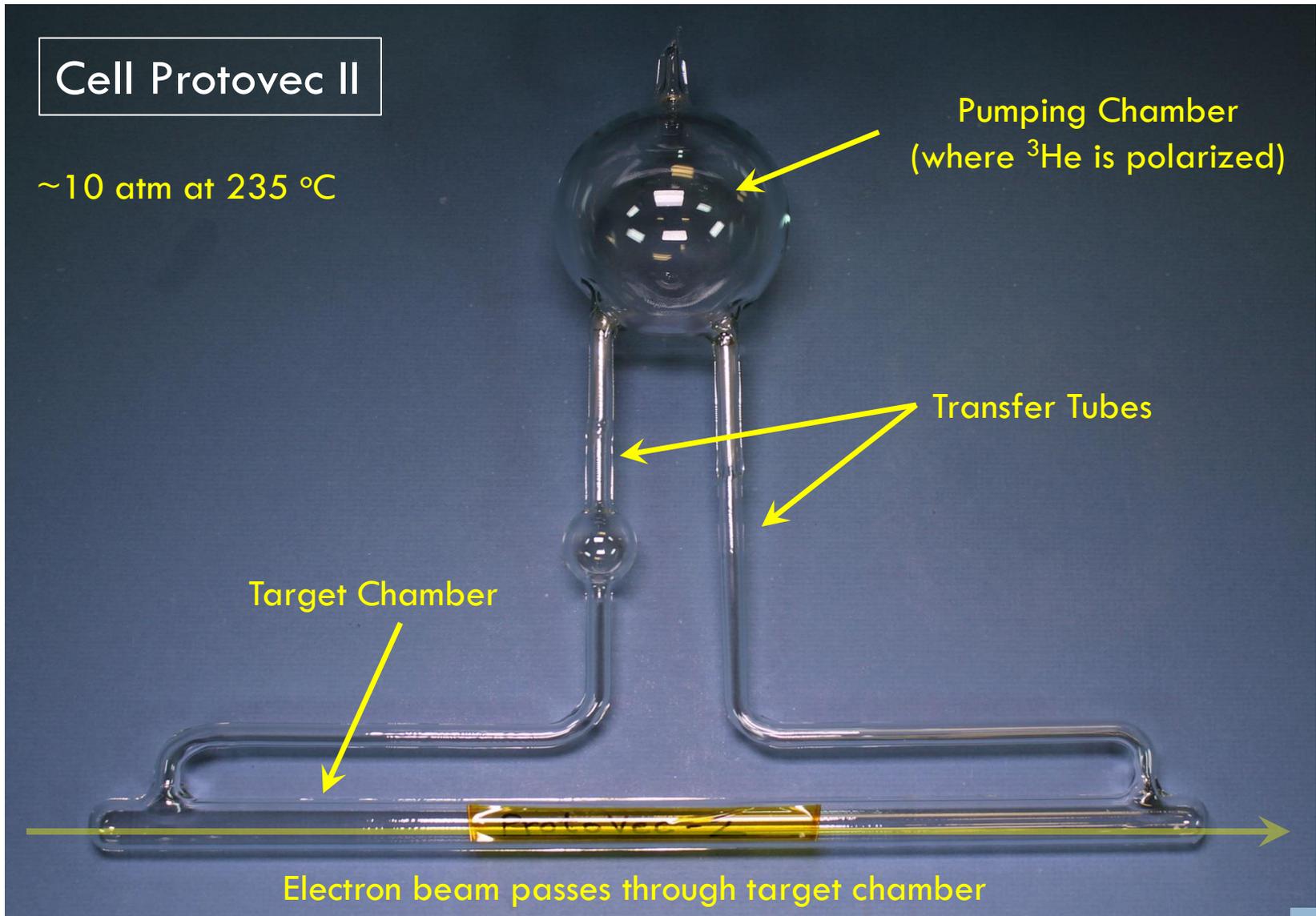
Adapted from P.A.M. Dolph et al., 2011

Ratio of polarizations of the target chamber to the pumping chamber quickly reaches unity with convection driven gas circulation.

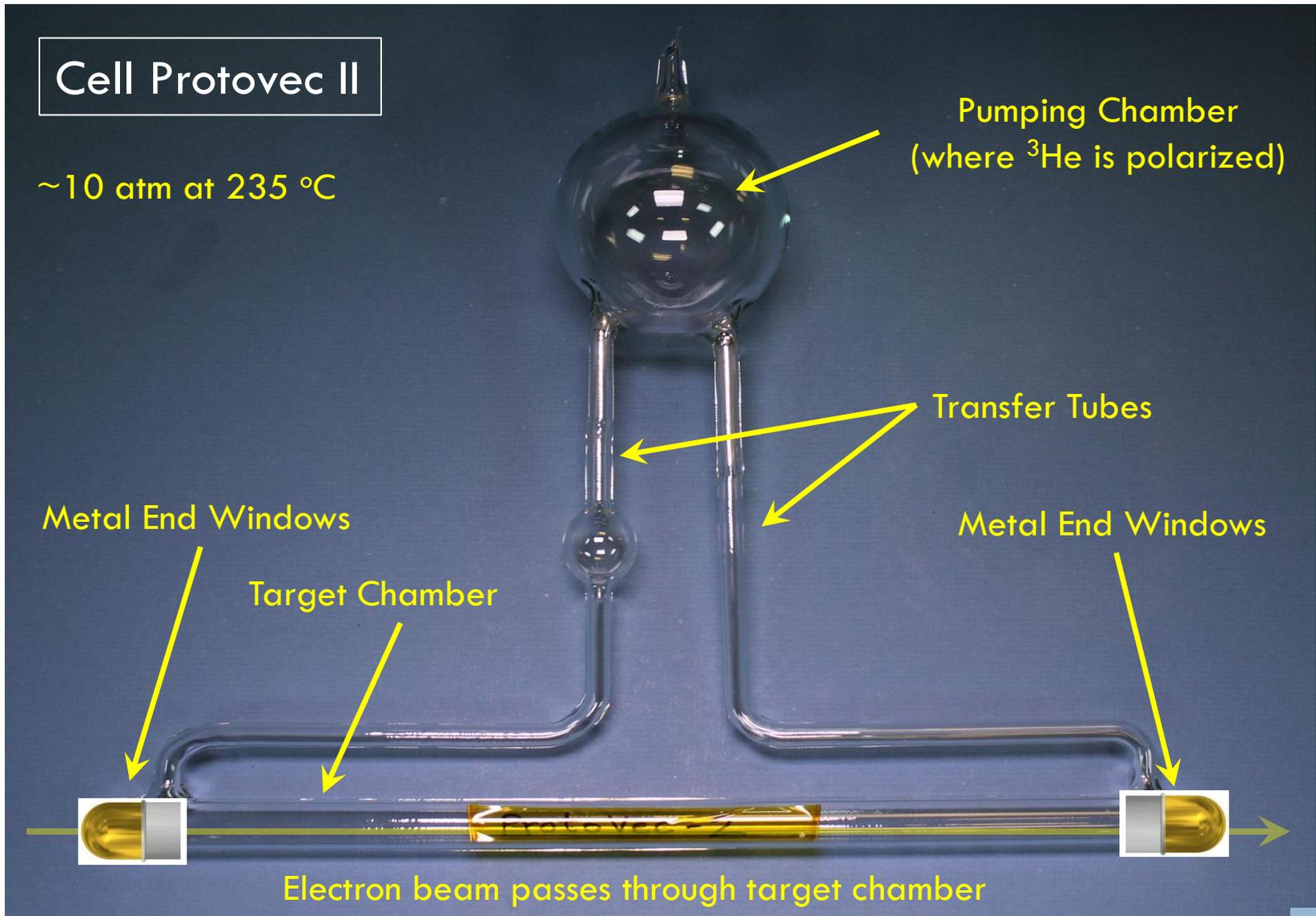
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Polarized ^3He Target



Polarized ^3He Target



Numbers on Gold from the literature look encouraging for making metal end windows.

Paramagnetic relaxation of spin polarized ^3He at coated glass walls

Part II

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Eur. Phys. J. D **38**, 439–443 (2006)
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5 Other coatings

In the course of 3 thesis works [18,23,27] we have briefly examined also a number of other coatings on various glass substrates and measured their ^3He relaxation times (in brackets): metals: Mg (6 h), Al (8 h), Zn (12 h), Se (5 h), Ag (5 h), Sb (7 h), Te (10 h), Au (20 h), Pb (26 h), Bi (50 h); salts: LiF (8 h), MgF₂ (8 h), CsF (25 h), CsCl (18 h); oxides: Al₂O₃ (4 h); hard covalent coatings: diamond (3 h), titanium nitride (2 h). Most of the more volatile species have been evaporated in situ by a removable oven and ^3He relaxation has been measured without breaking the vacuum in between [18]. This might have in-

Wall Relaxation Rate

$$1/T_1^{\text{wall}} = \rho A / V$$

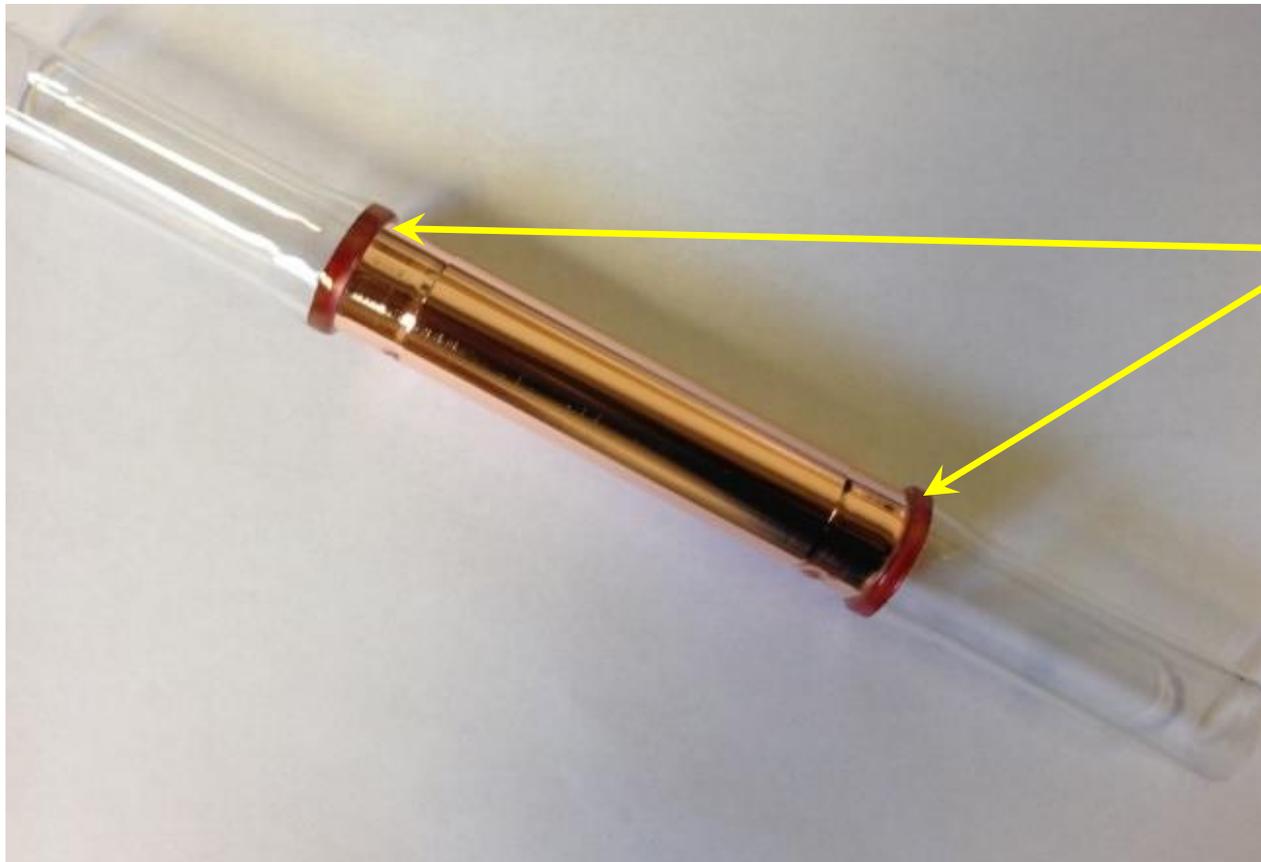
ρ - Surface relaxivity

A - Area

V - Total Volume

Preparation of Pyrex-Copper-Pyrex tubes

Larson Glass-Metal Seals



Housekeeper Seals

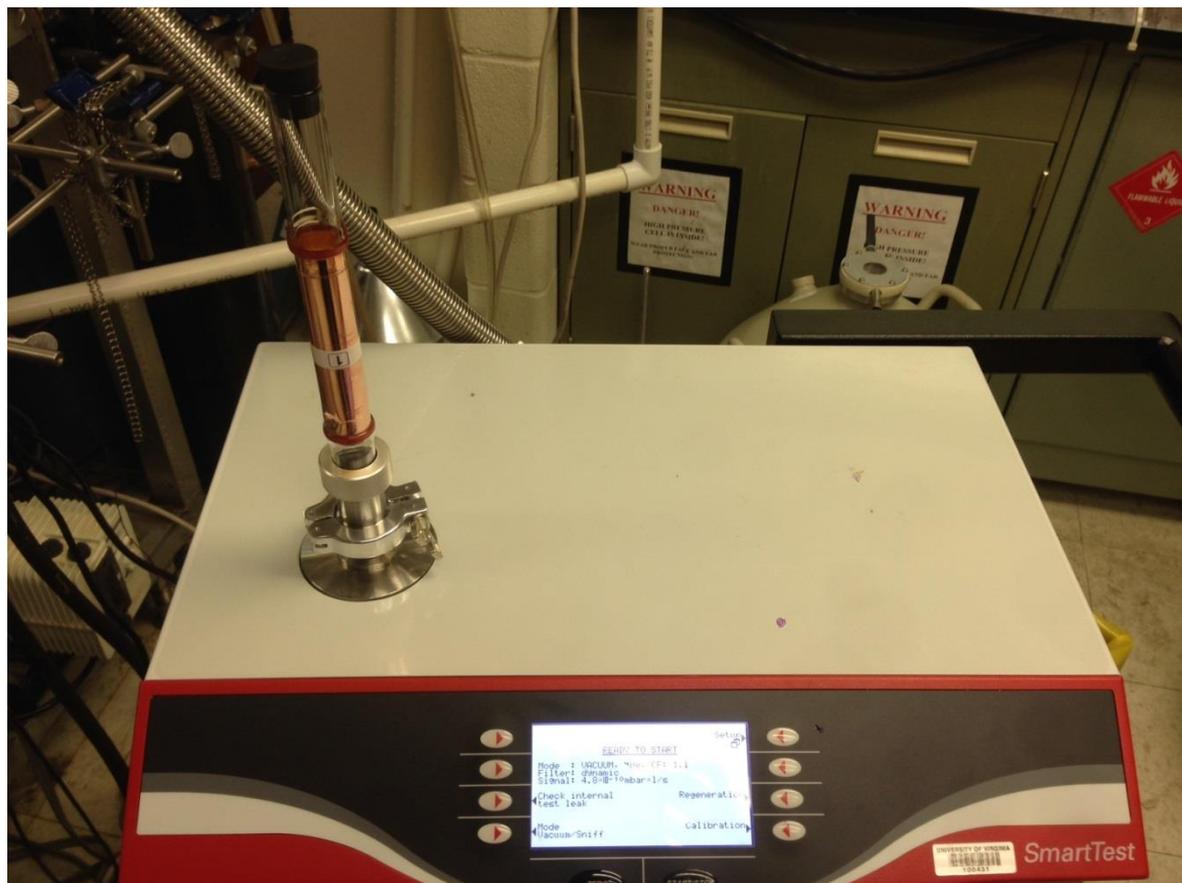
Preparation of Pyrex-Copper-Pyrex tubes

Pressure tested Glass-Metal Seal. It survived pressures greater than 20 atm.



Preparation of Pyrex-Copper-Pyrex tubes

Leak tested the seals. Leak Rate about 10^{-9} - 10^{-10} mbar l/s.



Preparation of Pyrex-Copper-Pyrex tubes

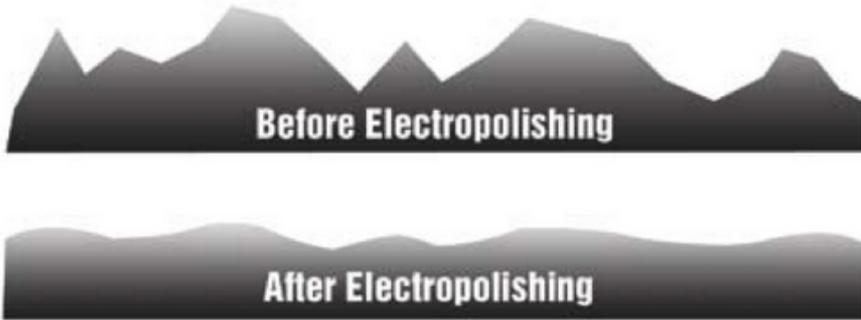
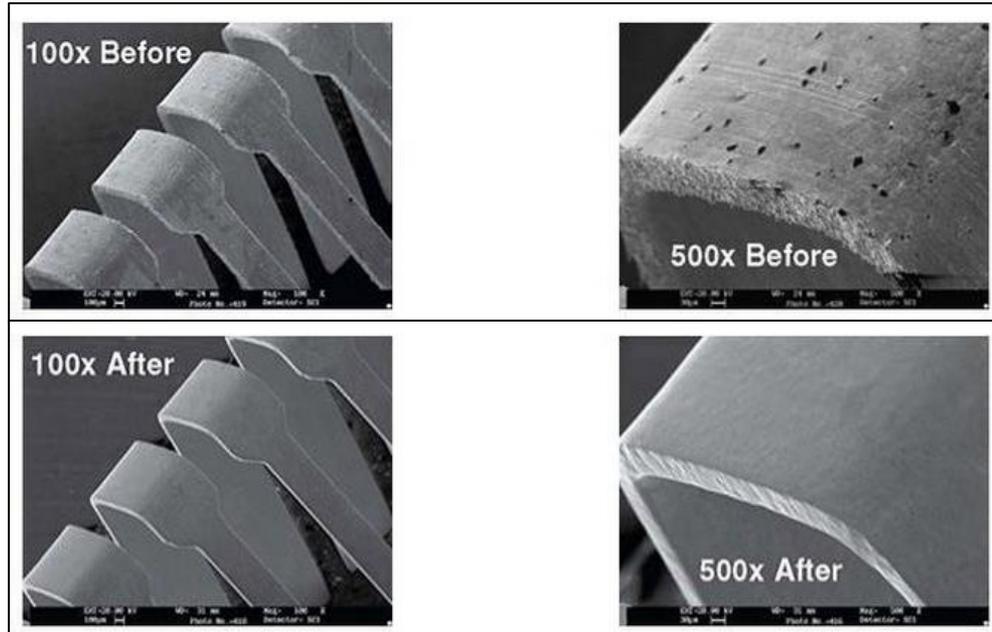


Illustration A

In general terms, when properly applied, electropolishing can reduce microfinish values by 50% with a removal of .0005" from each surface. Chart B clearly shows that maximum benefit is achieved in this area, and that removing much more metal does not continue to improve surface finish.



② Electro-polished



Preparation of Pyrex-Copper-Pyrex tubes



Laser Gold

Laser Gold plating is a proprietary, pure, hard, electrochemically deposited gold coating that combines the theoretical reflectivity and emissivity of gold, with a surface that can *actually be physically cleaned!*



A process that has been refined for more than twenty years, Laser Gold has been the sole and single NIST standard (#2011) for infrared reflective material. Epner Technology is also proud of the fact that Laser Gold has been specified on some of our nation's most advanced military and space programs.

<http://epner.com>

③ Gold plated inside surface (for some tubes)

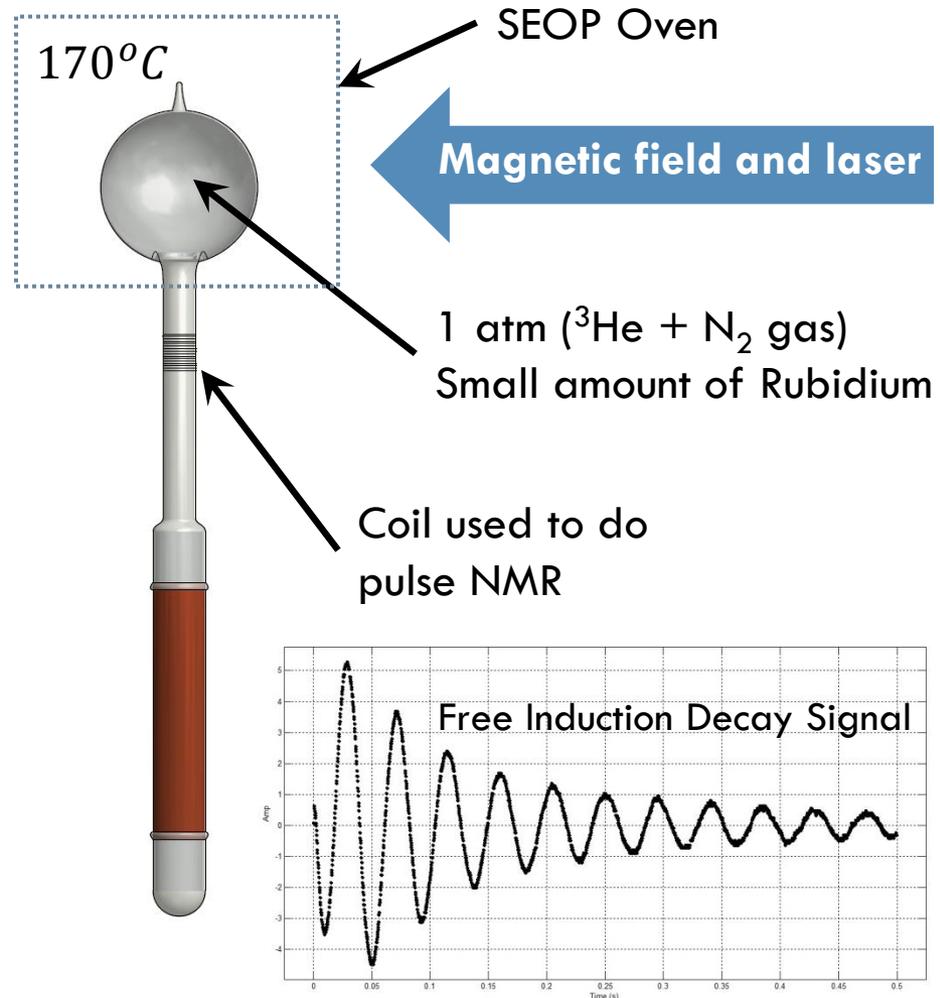
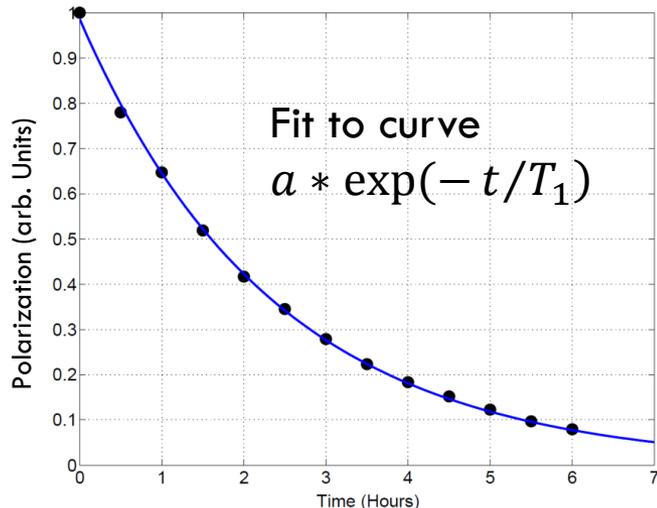
Preparation of Pyrex-Copper-Pyrex tubes



④ Cleaned tubes using
ultra-sonic cleaner.

Method for testing samples

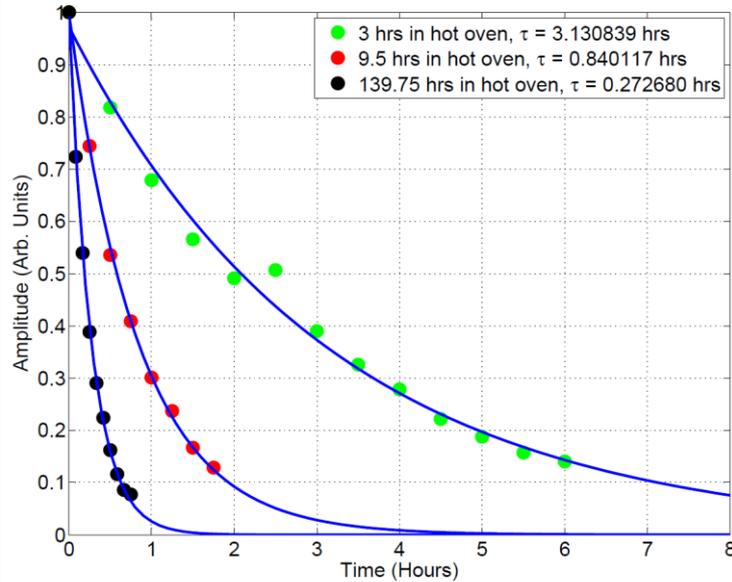
- ^3He test cells built using tubes
- ^3He polarized using spin-exchange optical pumping (SEOP)
- Polarization monitored using Pulse NMR
- Polarization measured as a function of time.



Results with and without Gold

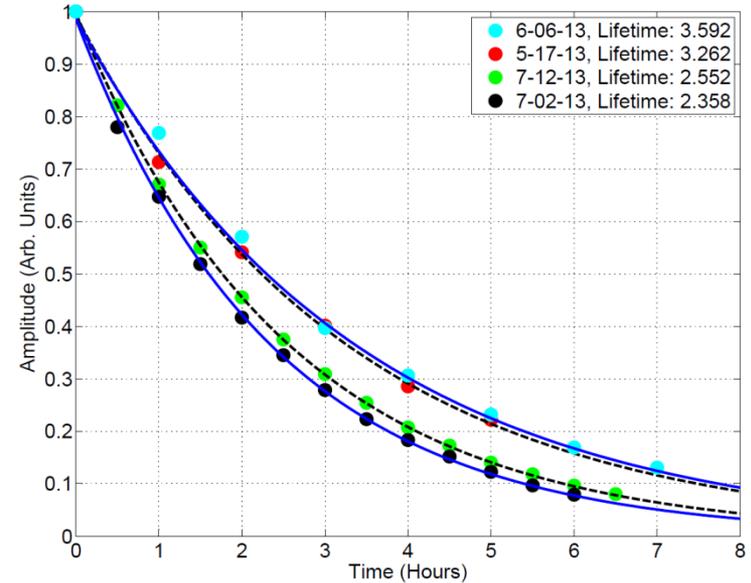


“Cupid” - Copper-only



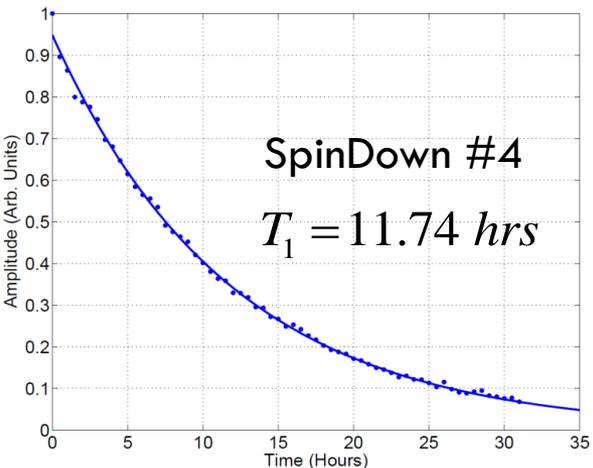
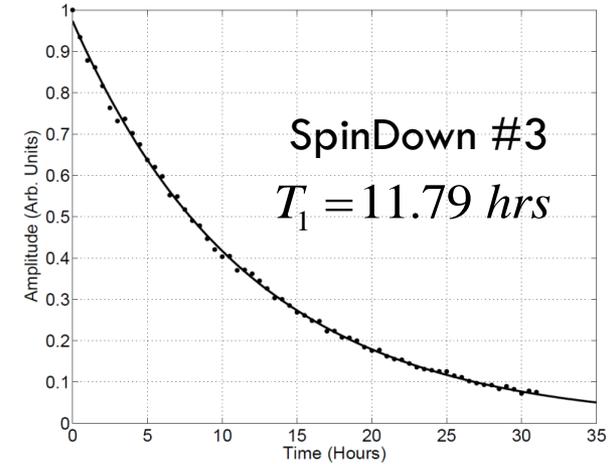
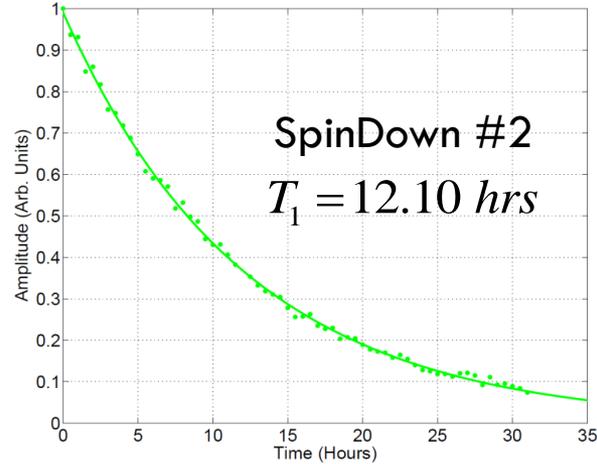
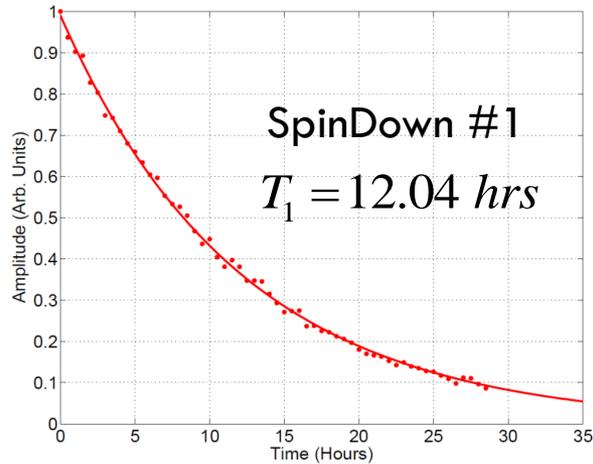
Lifetimes degrade from 3.1 hrs to 0.3 hrs. We believe Rb exposure degraded our surfaces.

“Goldfinger” - Gold-coated copper



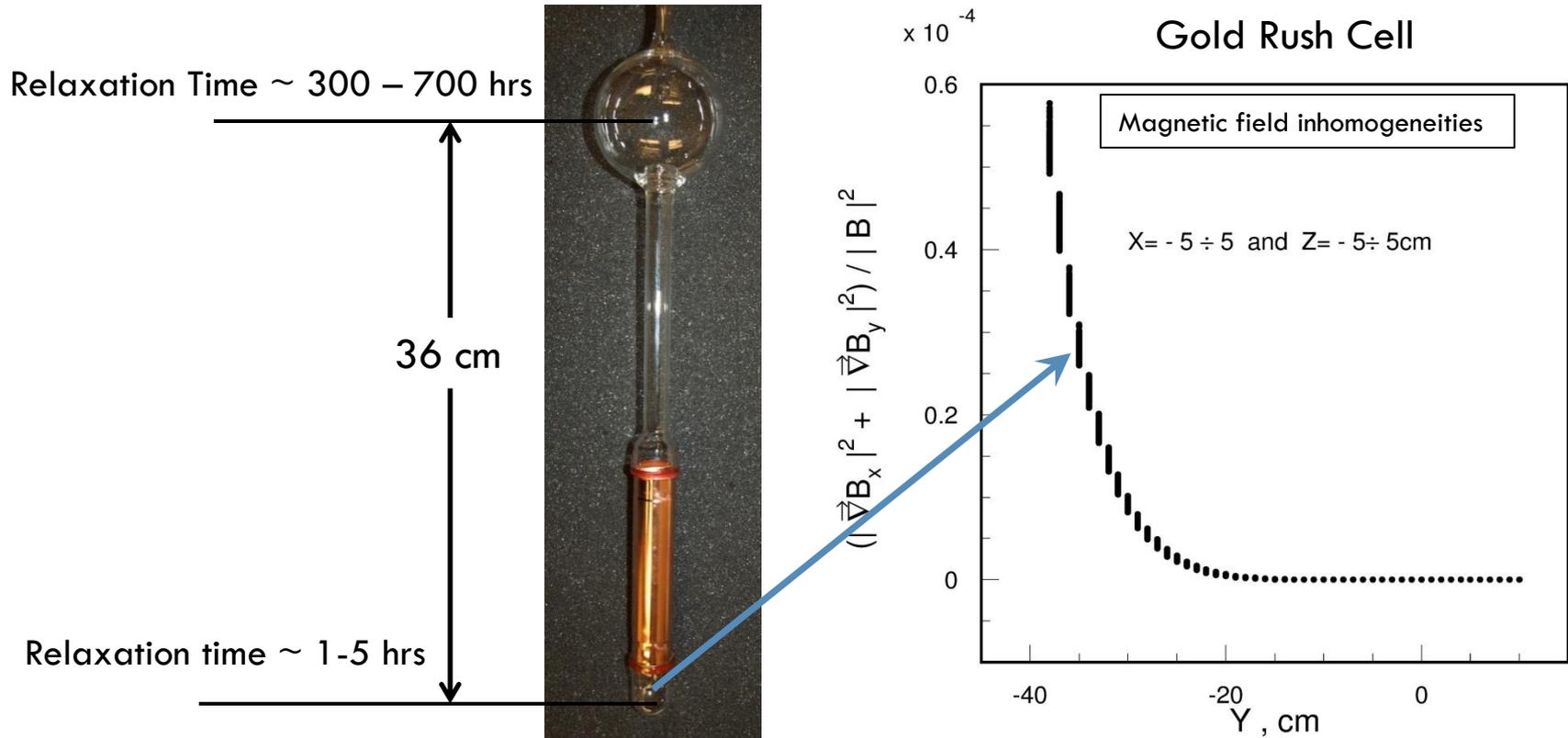
Lifetime degraded only slightly from 3.6 hrs to 2.4 hrs.

Tests of “GoldRush”



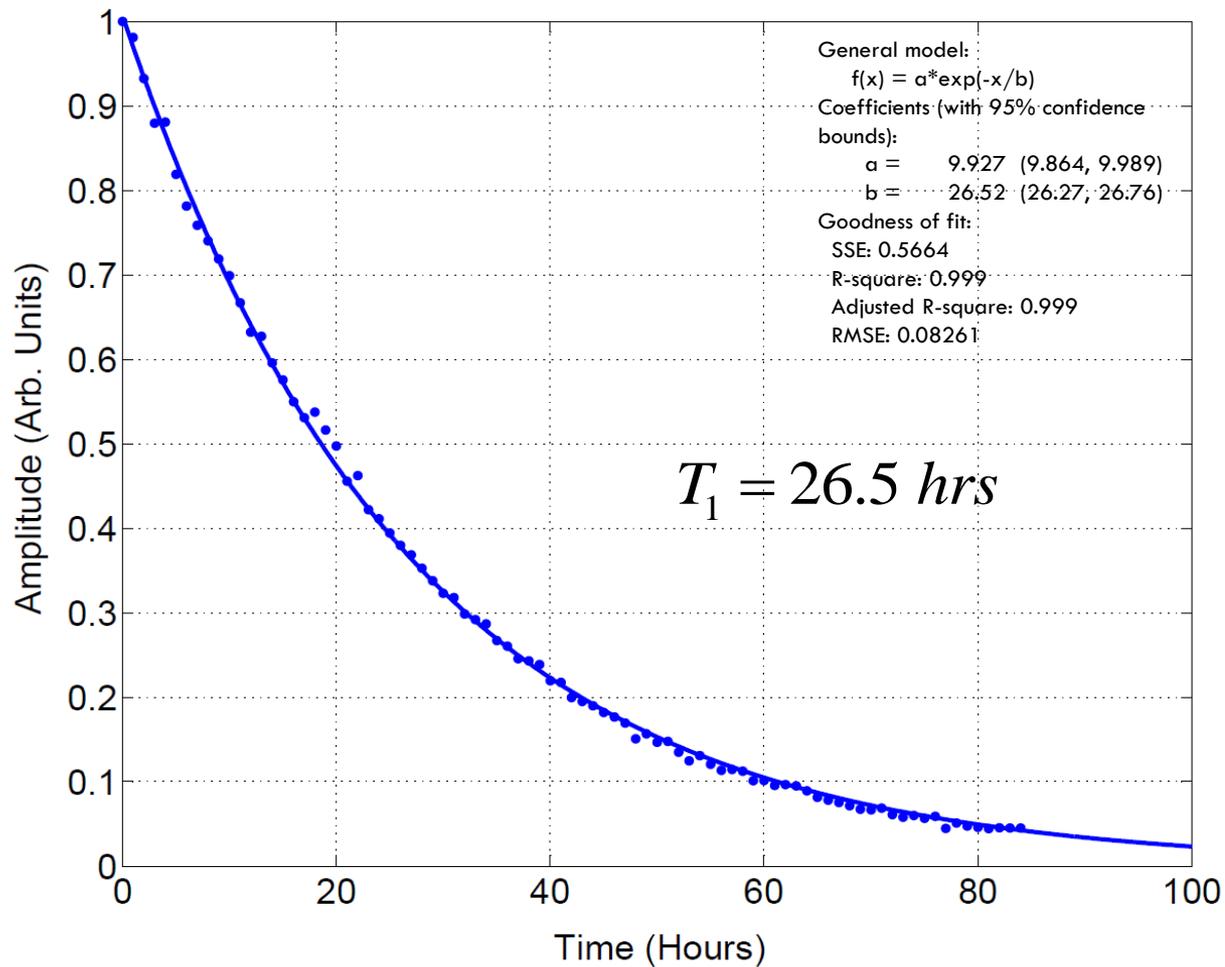
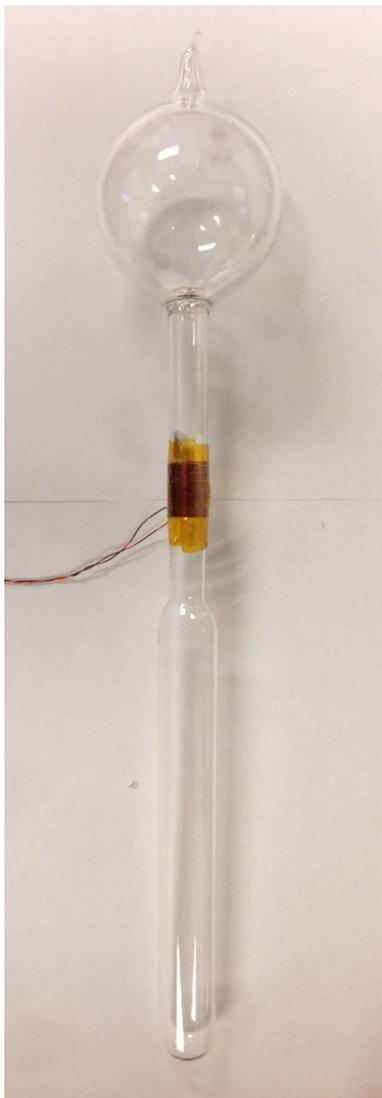
- a) No significant degradation of lifetime was observed over four spin downs
- b) The observed lifetimes were the longest we had measured to date.
- c) We further discovered these lifetimes were limited by magnetic field inhomogeneities (more on this to come)

Calculations indicated lifetime of “GoldRush” was at least partially limited by magnetic field inhomogeneities



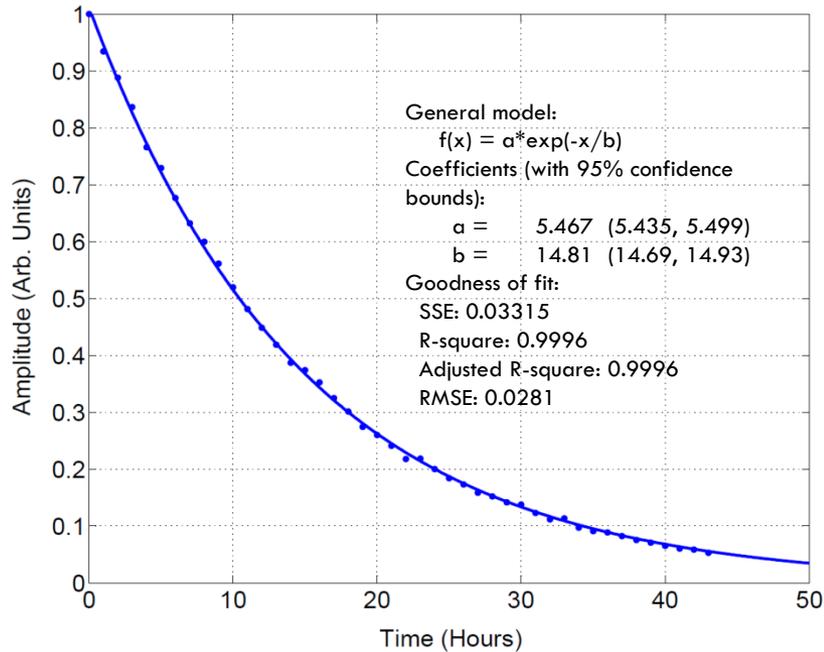
Lifting the cell by ~ 5 cm improved lifetime of GoldRush from around 11 to 15 hrs.

Control Cell "Pyrah" for glass-metal cell tests

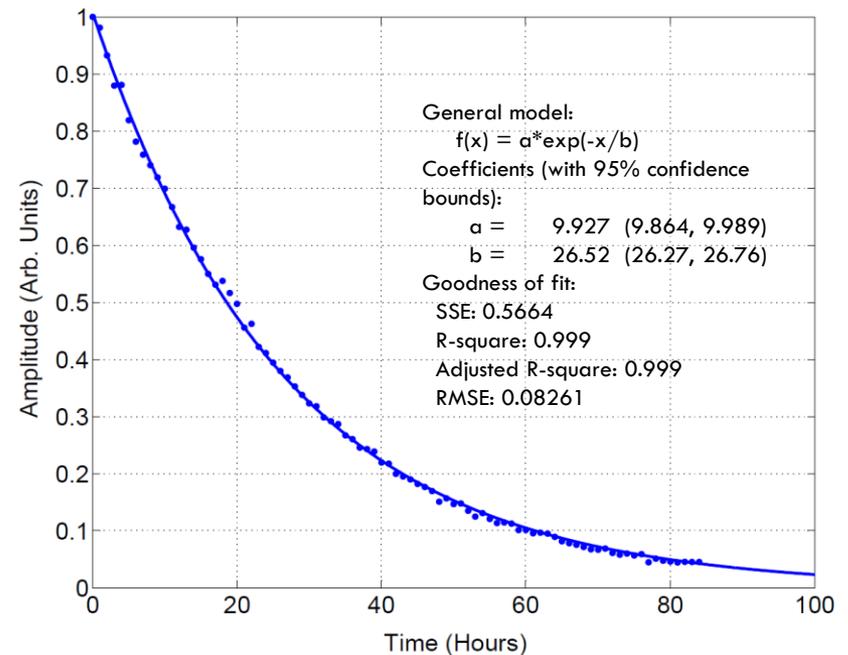


Cell Pyrah

“GoldRush” and “Pyrah” in Elevated position (lower average magnetic field inhomogeneities)



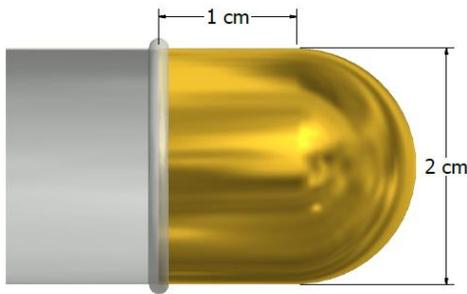
Gold Coated Cell $T_1 = 14.81 \text{ hrs}$



Control Cell $T_1 = 26.5 \text{ hrs}$

Comparing the control with Goldrush suggests the metal introduces additional relaxation (in this size cell) of $1/25 \text{ hrs}$!!!

Conclusion: Metal end caps appear feasible



OFHC Copper coated with gold appears to give excellent results.

Two end caps like this on Protovec would only contribute $\sim 1/135$ hrs for relaxation.



Currently Developing a target cell with these end windows.

Also exploring “Titanium” coated with gold or platinum

Summary

- Convection based cells reduce polarization gradients between pumping and target chambers.
- Convection based cells avoid radiation damage to pumping chamber due to electron beam.
- Targets with metal end windows allow us to use higher beam currents.

Acknowledgement

- Spin Physics Group – UVA

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Dr. William A. Tobias

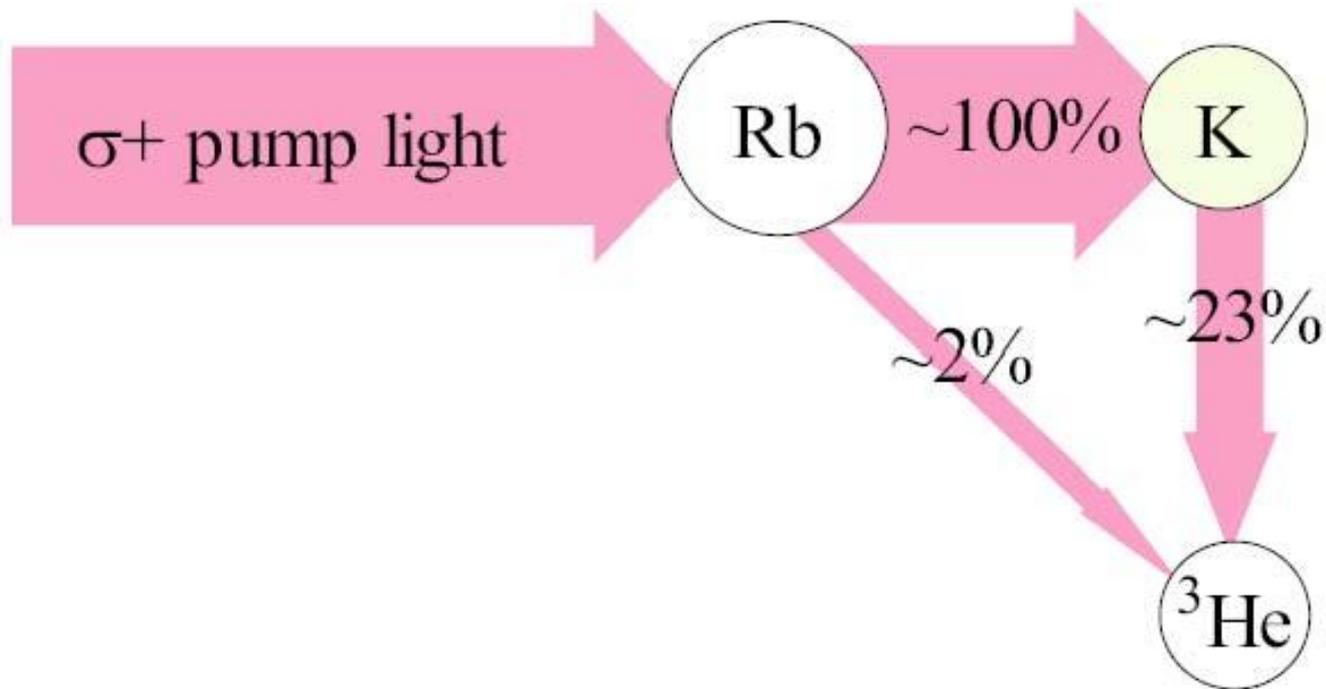
Dr. Vladimir Nelyubin

- Mike Souza, Univ. of Princeton – for making all the cells

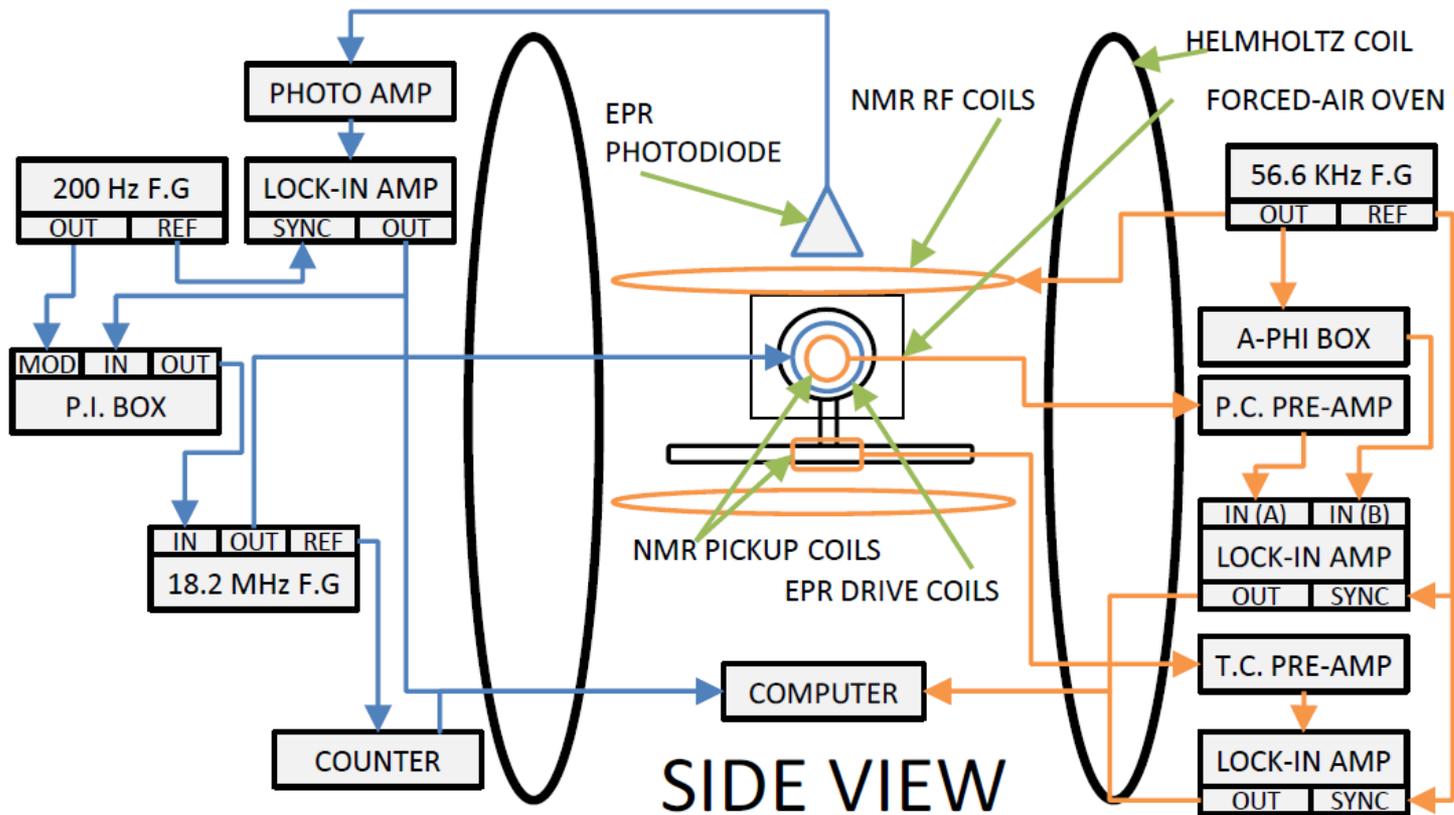


Backup Slides

Reason to use Alkali Hybrid Mixture



NMR and EPR setup



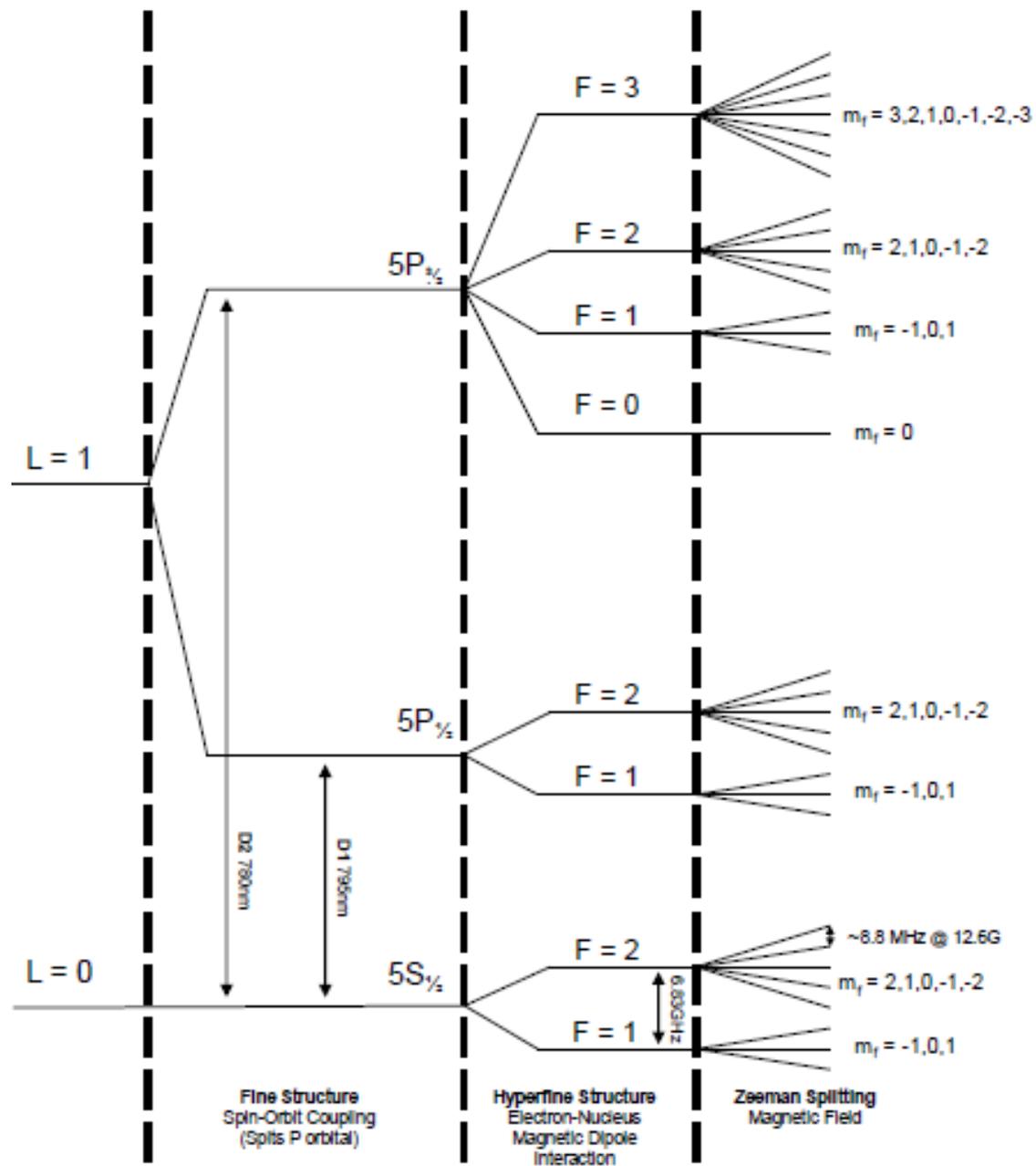


Figure 2.2: Energy-level diagram of ^{87}Rb , ($I = 3/2$, not to scale) showing full hyperfine structure

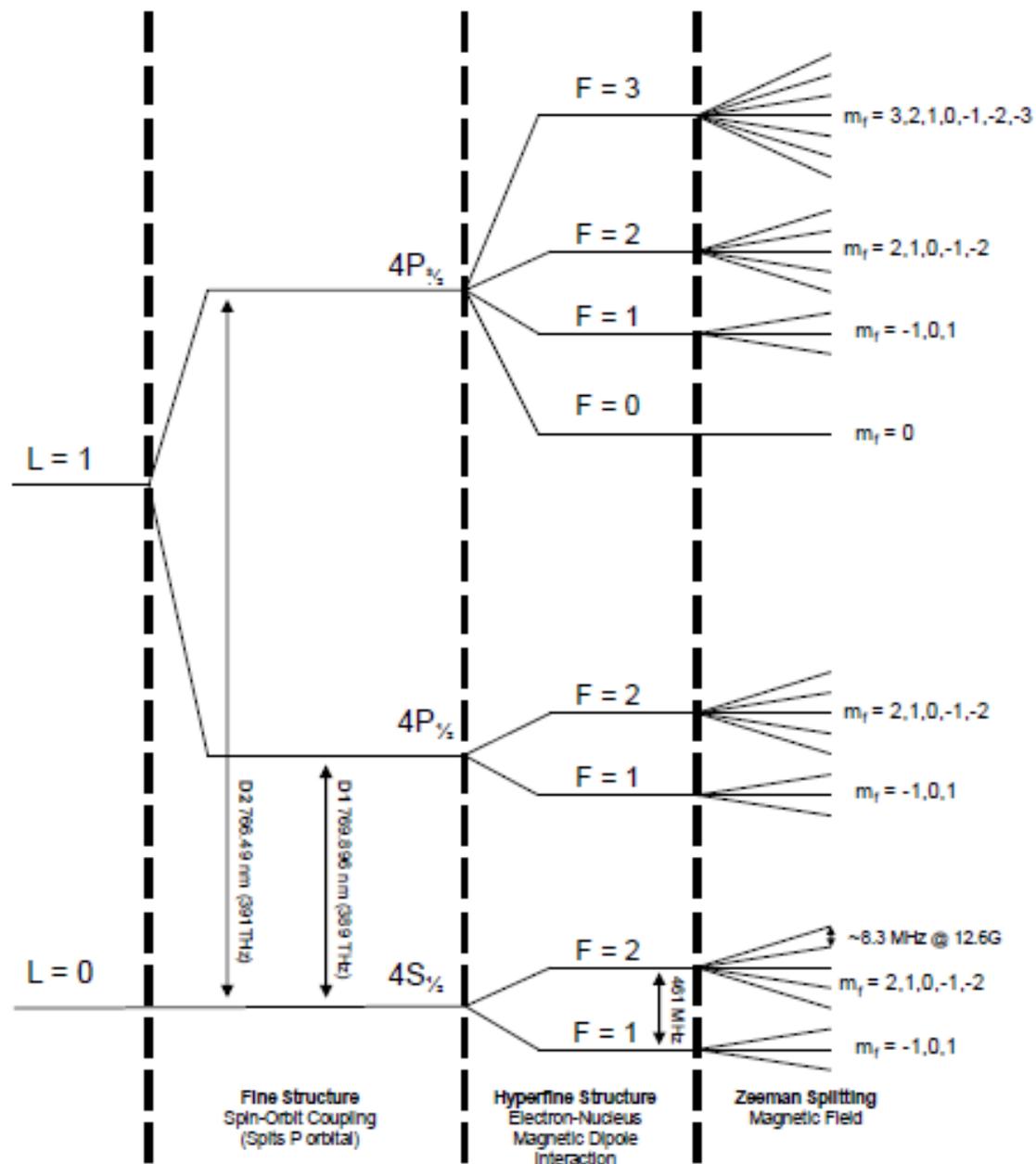


Figure 3.4: Energy-level diagram of ^{39}K , ($I = 3/2$, not to scale)